

HW 1.2 ch 9

13] closed system with four processes

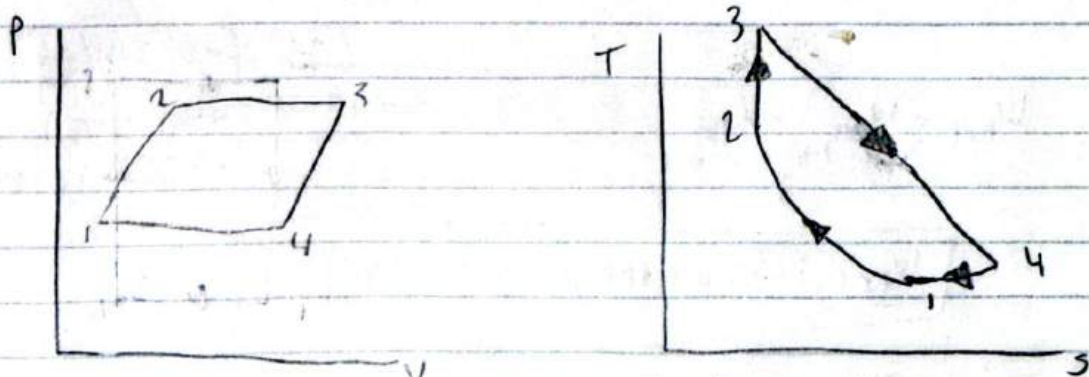
1-2 Isentropic compression from 100 kPa and 22 °C to 600 kPa

2-3 $V = \text{constant}$ heat addition to 1500 K

3-4 Isentropic expansion to 100 kPa

4-1 $P = \text{constant}$ heat rejection to initial

- show cycle on P - V and T - S diagram
- calculate net work output per unit mass
- determine thermal efficiency



isen thermat

V
1) 1000

serbische

$$\frac{V_4}{V_3} = \frac{T_4}{T_3}$$

①

②

③

(4)

$$V = \frac{(.257)(295)}{100}$$

$$y = \frac{(.287)(1700)}{500}$$

 ρ_{const}

$$P = \frac{(-.207) / (1500)}{.84665}$$

$$PV = RT$$

$$T_4 = \frac{P_4}{P_3} T_3 =$$

$$T_4 = \frac{100}{508.47} \text{ (1)}$$

$$V_4 = \frac{295}{1500} \text{ l. by}$$

$$P_1 = 100 \mu P_0$$

$$P_2 = 600 \text{ kPa}$$

$$P_3 = 508.47$$

$$P_y = 100 \text{ kPa}$$

$$T_1 = 22^\circ\text{C} = 295\text{K}$$

$$T_2 = \frac{r_2}{r_1} T_1$$

$$T_3 = 1500 \text{ K}$$

$$T_4 = 295.00$$

$$v_1 = .84665$$

$$T_1 = 1770$$

$V_2 = .84665$

$$v_4 = .1665$$

11 = 64665

$$W_{in\ 4-1} = P_4(V_4 - V_1) = 100(1.665 - 0.84665) = -68.015 \frac{kJ}{kg}$$

$$W_{out\ 2-3}$$

$$(q_{in\ 2-3} - \cancel{q_{out}}_0) + (\cancel{w_{in}}_0 - w_{out\ 1-2}) = U_3 - U_2$$

$$q_{in\ 2-3} - w_{out\ 2-3} = C_v(T_3 - T_2) \quad \text{Iso}$$

$$\frac{w_{comp}}{w_{exp}} = \frac{-68.015}{87.159}$$

$$w_{out\ 2-3} = q_{in\ 2-3}$$

$$w_{out\ 2-3} = T_3(s_3 - s_2)$$

$$\frac{w_{comp}}{w_{exp}} = -0.78$$

$$w_{out\ 2-3} = RT_3 \ln\left(\frac{V_3}{V_2}\right)$$

$$(1.287)(1500) \ln\left(\frac{\sqrt{3}}{\sqrt{2}}\right)$$

$$430(1.1) \ln\left(\frac{\sqrt{3}}{\sqrt{2}}\right)$$

$$(430) \ln(1.2247)$$

$$w_{net} = w_{out\ 2-3} - w_{in\ 4-1}$$

$$w_{net} = 87.159 + 68.015 = 155.174 \frac{kJ}{kg}$$

$$w_{out\ 2-3} = 87.159$$

$$\eta_{th} = \frac{w_{net}}{q_{in}}$$

$$q_{in} = q_{in\ 1-2} + q_{in\ 2-3}$$

$$q_{in} = 1146.209$$

$$q_{in\ 1-2} = 0.718(1770 - 295)$$

$$\frac{155.174}{1146.209} = 0.1354$$

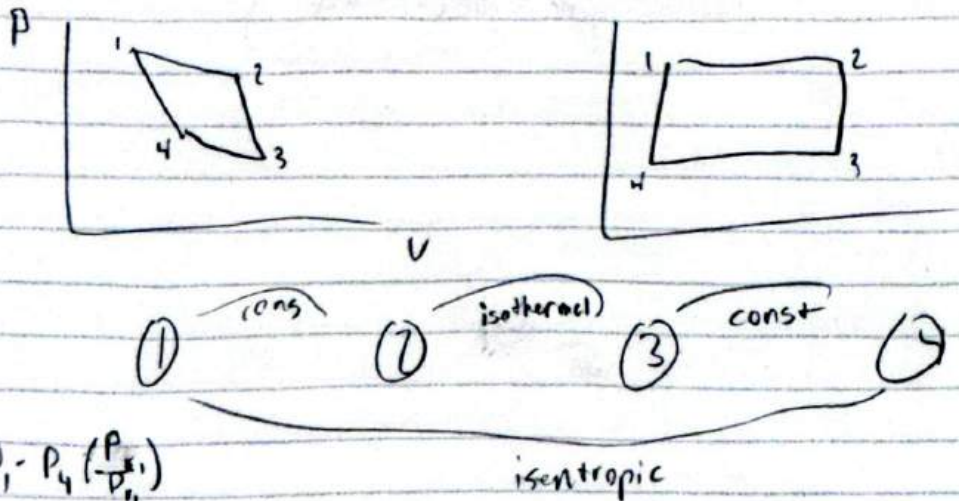
$$q_{in\ 1-2} = 1059.0$$

$$\eta_{th} = 13.54\%$$

Carnot cycle

9-19

temp limits 350k and 1200k. pressure before and after isothermal compression are 150 and 300 kPa. net work per cycle is .5 kJ determine (a) max pressure (b) heat transfer to air and (c) mass of air



$$P_1 = P_4 \left(\frac{P_2}{P_1} \right)$$

$$300 \left(\frac{230}{2.279} \right)$$

$$P_1 = 30.01 \text{ MPa} = P_{\text{max}}$$

$$\eta_c = 1 - \frac{T_L}{T_H}$$

$$1 - \left(\frac{350}{1200} \right)$$

$$\eta_c = .7083$$

$$\eta_c = \frac{W_{\text{out}}}{Q_s} = .7083 = \frac{.5}{Q_s}$$

$$Q_{\text{out}} = .706 \text{ kJ}$$

$$ds = \frac{dQ}{T} = dQ = T ds$$

$$dW = (T_H - T_L)(s_4 - s_2)$$

$$(s_4 - s_3) = -(s_3 - s_4)$$

$$R \ln \left(\frac{P_4}{P_3} \right)$$

$$.2070 \ln \left(\frac{300}{150} \right)$$

$$.1969 \text{ kJ/kg} \cdot \text{K}$$

$$dW = (T_H - T_L)(s_4 - s_3)$$

$$(1200 - 350)(.1969)$$

$$dW = 169.065 \text{ kJ/kg}$$

$$m = \frac{w}{dv} = \frac{.5}{169.065}$$

↓

$$m = .002957 \text{ kg}$$

1-2 isentropic compression initial temp $T_1 = 70^\circ\text{C}$ with compression ratio $r = 5$

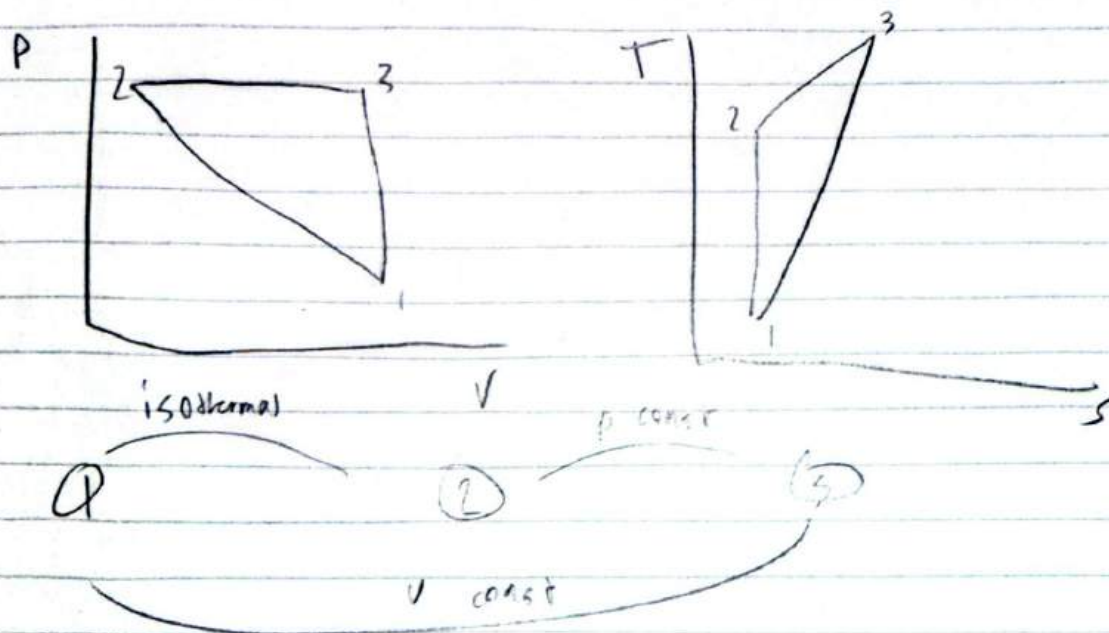
2-3 constant pressure heat addition

$$C_v = .7 \text{ kJ/kg}\cdot\text{K}$$

3-1 constant volume heat rejection

$$R = .3 \text{ kJ/kg}\cdot\text{K}$$

- sketch diagrams
- heat work interactions
- thermal efficiency
- expression for cycle thermal efficiency as function of the compression ratio r and ratio of specific heats γ



$$1-2 \quad T_2 = T_1 (r^{k-1})$$

$$293 (5^{1.42857-1})$$

$$T_2 = 584.02 \text{ K}$$

$$q_{1-2} = 0 \text{ kJ/kg}$$

$$w_{1-2} = \frac{R(T_1 - T_2)}{k-1}$$

$$= \frac{3(293 - 584.02)}{1.42857-1}$$

$$w_{1-2} = -203.72 \text{ kJ/kg}$$

$$2-3 \quad \frac{V_2}{T_2} = \frac{V_3}{T_3}$$

$$T_3 = T_2 \left(\frac{V_3}{V_2} \right) = T_2 \left(\frac{V_1}{V_2} \right) = T_2 \cdot r$$

$$\downarrow$$

$$584.02(5)$$

$$T_3 = 2920.1 \text{ K}$$

$$q_{2-3} = c_p(T_3 - T_2)$$

$$1(2920.1 - 584.02)$$

$$q_{2-3} = 2336.08$$

$$\Delta u_{2-3} = u_3 - u_2 = (u(T_3 - T_2))$$

$$.7(2920.1 - 584.02)$$

$$1633.26$$

$$w_{2-3} = q_{2-3} - \Delta u_{2-3}$$

$$w_{2-3} = 700,82 \text{ kJ/kg}$$

3-1

$$q_{3-1} = c_v (T_1 - T_3)$$

$$.7(293 - 2920.1)$$

$$q_{3-1} = -1838.97 \text{ kJ/kg}$$

$$w_{3-1} = 0 \text{ kJ/kg}$$

$$\eta_{TH} = 1 - \frac{q_{out}}{q_{in}}$$

$$1 - \frac{1838.97 (100)}{2536.69}$$

$$\downarrow$$

$$.2176 = 21.76\%$$

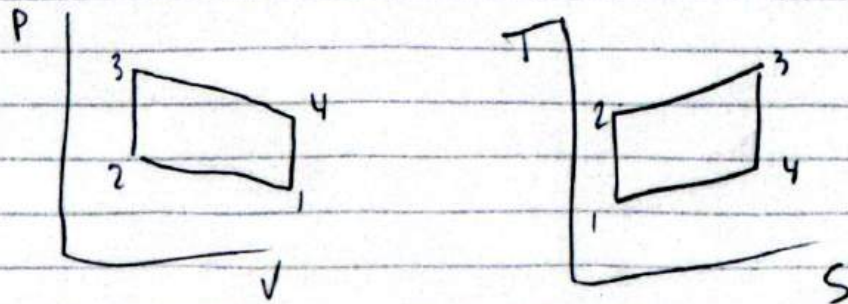
$$\eta_{TH} = 1 - \frac{1}{k} \cdot \frac{1}{r^{k-1}} \cdot \frac{r^{k-1}}{r-1}$$

$$\eta_{TH} = 1 - \frac{1}{r^{k-1}}$$

9-31

Otto cycle compression ratio of 10.5 takes in air at 90 kPa and 40°C and is repeated 2500 times per min

Determine thermal efficiency and rate of heat input if cycle is to produce 90 kW of power



$$k = \frac{C_p}{C_v} \quad k = \frac{1.005}{.718} = 1.399$$

$$\eta = 1 - \frac{1}{r^{k-1}}$$

$$1 - \frac{1}{(10.5)^{1.399-1}} = 1 - \frac{1}{2.555} = 1 - .391 = .609$$

$$\eta = 60.9\%$$

$$ROH = P / \text{cycle frequency}$$

$$ROH = \frac{90000}{150000}$$

$$\begin{array}{c} 90 \text{ kW} \\ \downarrow \\ 90000 \end{array}$$

$$\begin{array}{c} 2500 \\ \downarrow \\ 150000 \text{ cycles} \end{array}$$

$$ROH = .6$$