

3.11 If you were to ride in an open-cockpit airplane to an elevation of 4000 ft above sea level, what would the atmospheric pressure be if it conforms to the standard atmosphere?

$$h = 4000 \text{ ft}$$

$$\Delta p = \gamma h$$

$$p_{\text{atm}} = 14.70 \text{ psia}$$

$$\gamma = 0.0764 \text{ lb/ft}^3$$

$$\Delta p = (0.0764 \text{ lb/ft}^3)(4000 \text{ ft})$$

$$\Delta p = 305.6 \left(\frac{\text{lb}}{(\text{ft})^2} \right)$$

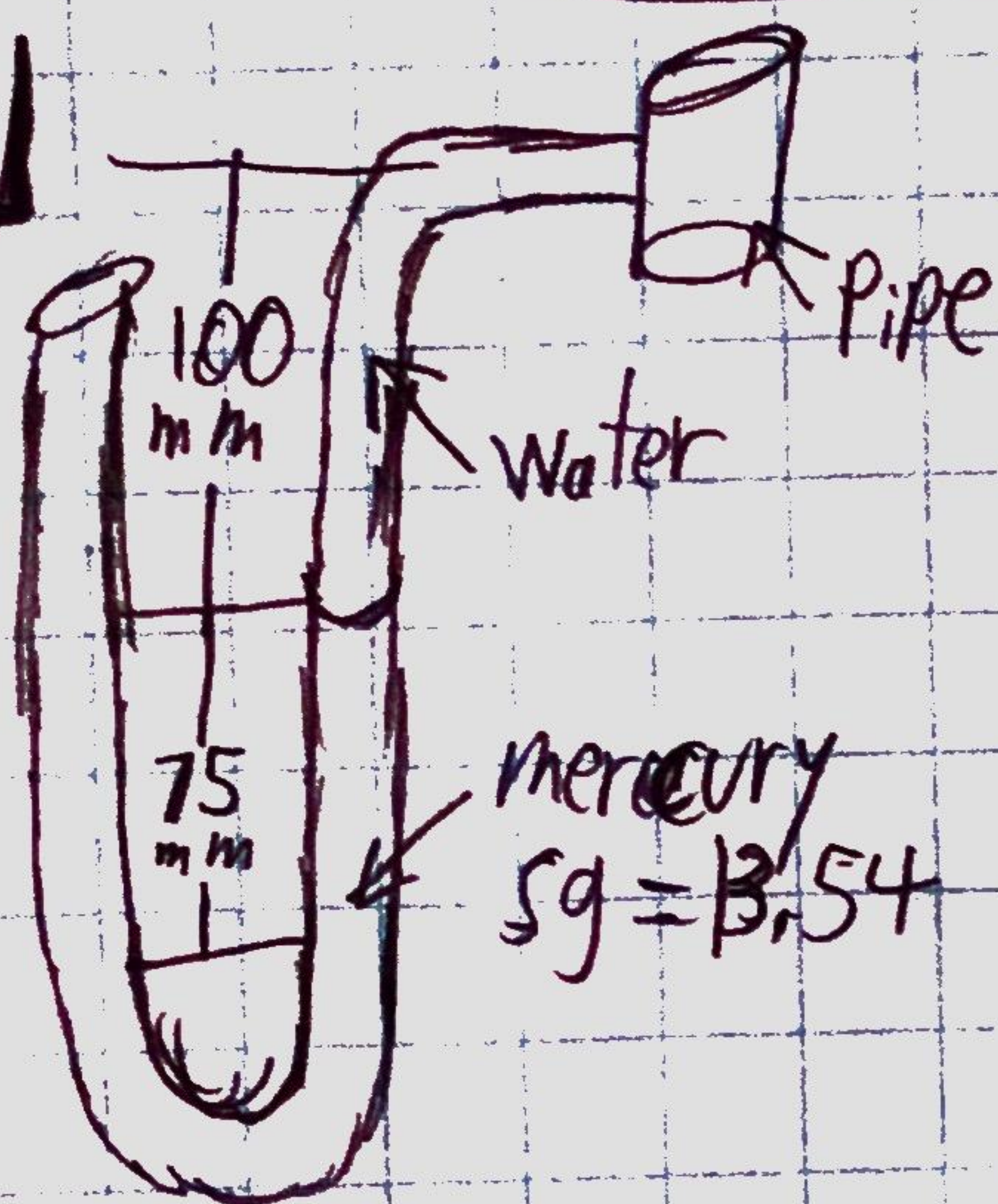
$$\Delta p = 2.12 \text{ psia}$$

$$p = p_{\text{atm}} - \Delta p$$

$$p = 14.70 \text{ psia} - 2.12 \text{ psia}$$

$$p = 12.58 \text{ psia}$$

3.62



$$p_A + \gamma_w h_w + \gamma_{Hg} h_{Hg} = p_{\text{atm}}$$

$$p_A - p_{\text{atm}} = -(\gamma_w h_w + \gamma_{Hg} h_{Hg})$$

$$(p_A)_{\text{gage}} = -(\gamma_w h_w + \gamma_{Hg} h_{Hg})$$

$$(p_A)_{\text{gage}} = -(\gamma_w h_w + \gamma_w sg_{Hg} h_{Hg})$$

$$(p_A)_{\text{gage}} = -(9810 \times 0.1 + 9810 \times 13.54 \times 0.075)$$

$$= -10943.055 \text{ N/m}^2$$

$$= -10943.055 \text{ Pa}$$

$$= -10.943 \text{ kPa}$$

3.83

A barometer indicates the atmospheric pressure to be 30.65 in of mercury. Calculate the atmospheric pressure in psia.

$$p_{\text{atm}} = 30.65 \text{ in of Hg}$$

$$30.65 \text{ in} \times \frac{0.49 \text{ psia}}{1 \text{ in}}$$

$$= 15 \text{ psia}$$