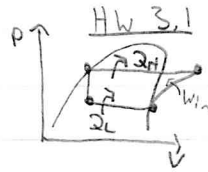
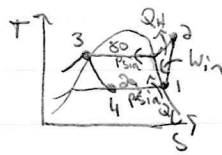


3/31/19

⑪



Sat. Refrigerant - 134a table

$$s_1 = s_2$$

$$h_1(h_g) = 102.74 \text{ Btu/lbm}$$

$$s_1(s_g) = 0.22570 \text{ Btu/lbm} \cdot R$$

$$h_2 = h_{f,s} + (h_{g,s} - h_{f,s}) \left( \frac{s_2 - s_{f,s}}{s_{g,s} - s_{f,s}} \right)$$

$$h_2 = 112.22 + (115.51 - 112.22) \left( \frac{0.22570 - 0.22545}{0.2263 - 0.22545} \right)$$

Sat. Refrigerant - 134a press. table

$$h_2 = 115.21 \text{ Btu/lbm}$$

$$h_3(h_f) = 33.39 \text{ Btu/lbm}$$

$$\dot{Q}_L = \dot{m}_{ice} (\Delta h)_{ice}$$

$$= \left( \frac{15}{3600} \right) (169)$$

$$\boxed{\dot{Q}_L = 0.7042 \text{ Btu/s}}$$

$$\dot{m}_g = \frac{\dot{Q}_L}{h_1 - h_4}$$

$$= \frac{0.7042}{(102.74 - 33.39)}$$

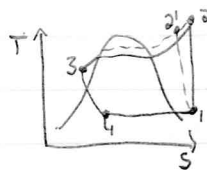
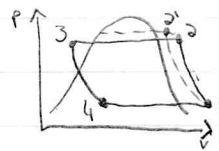
$$\boxed{\dot{m}_g = 0.0105 \text{ lbm/s}}$$

$$\dot{W}_{in} = \dot{m}_g (h_2 - h_1)$$

$$= (0.0105) (115.21 - 102.74) \left( \frac{1 \text{ hp}}{0.7068 \text{ Btu/s}} \right)$$

$$\boxed{\dot{W}_{in} = 0.176 \text{ hp}}$$

⑫



$$h_1 = 244.54 + \frac{(-10)(-5)(253.25 - 244.54)}{(-10 - 5)} \left. \vphantom{\frac{(-10)(-5)(253.25 - 244.54)}{(-10 - 5)}} \right\} \text{R-134a Table}$$

$$h_1 = 246.795 \text{ kJ/kg}$$

$$s_2 = 0.9360 + \frac{(10)(-5)(2.9698 - 0.9360)}{(-10 - 5)} \left. \vphantom{\frac{(10)(-5)(2.9698 - 0.9360)}{(-10 - 5)}} \right\} \text{R-134a Table}$$

Superheated R-134a Table

$$h_2 = 300.61 \text{ kJ/kg}$$

Sat. R-134a Table

$$h_3(h_f) = 114.28 \text{ kJ/kg}$$

$$h_4 = h_3$$

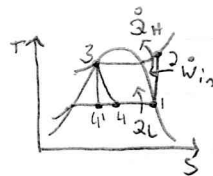
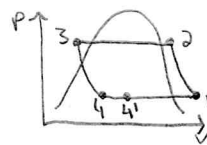
$$\begin{aligned} a) \dot{Q}_L &= \dot{m} (h_1 - h_4) \\ &= (0.07) (248.795 - 114.28) \\ &= 9.41 \frac{\text{kJ}}{\text{s}} \left( \frac{1 \text{ kW}}{1 \text{ kJ/s}} \right) \\ \dot{Q}_L &= 9.41 \text{ kW} \end{aligned}$$

$$\begin{aligned} \dot{W}_{in} &= \dot{m} (h_2 - h_1) \\ &= 0.07 (300.61 - 248.795) \\ &= 3.627 \frac{\text{kJ}}{\text{s}} \left( \frac{1 \text{ kW}}{1 \text{ kJ/s}} \right) \\ \dot{W}_{in} &= 3.627 \text{ kW} \end{aligned}$$

$$\begin{aligned} b) S_1 &= S_{2s} \quad S_1 = 0.9539 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \\ h_{2s} &= 278.27 + \frac{(0.9539 - 0.9297)(289.64 - 278.27)}{(0.9614 - 0.9297)} \quad \left. \begin{array}{l} \text{Superheated} \\ \text{R-134a tables} \end{array} \right\} \\ h_{2s} &= 286.199 \frac{\text{kJ}}{\text{kg}} \\ \eta_c &= \frac{(286.199 - 248.795)}{(300.61 - 248.795)} \\ &= 0.721 \\ \eta_c &= 72.1\% \end{aligned}$$

$$\begin{aligned} c) (COP)_R &= \frac{\dot{Q}_L}{\dot{W}_{in}} \\ &= \frac{9.41}{3.627} \\ (COP)_R &= 2.594 \end{aligned}$$

21)



Sat. R-134a Table

$$P_4 = P_1$$

$$h_f = 22.49 \text{ kJ/kg}$$

$$h_1(h_g) = 236.96 \text{ kJ/kg}$$

$$T_1(T_{sat}) = -22.32^\circ\text{C}$$

$$\begin{aligned} h_4 &= h_f + x_4 (h_g - h_f) \\ &= 22.49 + 0.34 (236.96 - 22.49) \\ h_4 &= 95.41 \text{ kJ/kg} \end{aligned}$$

$$b) h_3 = h_4$$

$$P_{3, \text{sat}} = 800 \text{ kPa} \quad \begin{array}{l} \text{Sat. R-134a} \\ \text{Press. Table} \end{array}$$

$$a) h_2 = 326.86 \text{ kJ/kg} \quad \begin{array}{l} \text{Super heated} \\ \text{R-134a Table} \end{array}$$

$$\dot{W}_{\text{comp}} = \dot{m}(h_2 - h_1)$$

$$450 = \dot{m}(326.86 - 236.97) \times 10^3$$

$$\dot{m} = 0.00644 \text{ kg/s}$$

$$c) \dot{Q}_{\text{evap}} = \dot{m}(h_1 - h_4)$$

$$= 0.00644(236.97 - 95.41) \times 10^3$$

$$\dot{Q}_{\text{evap}} = 911.56 \text{ W}$$

$$\text{COP} = \frac{\dot{Q}_{\text{evap}}}{\dot{W}_{\text{comp}}} = \frac{911.56}{450}$$

$$\text{COP} = 2.026$$



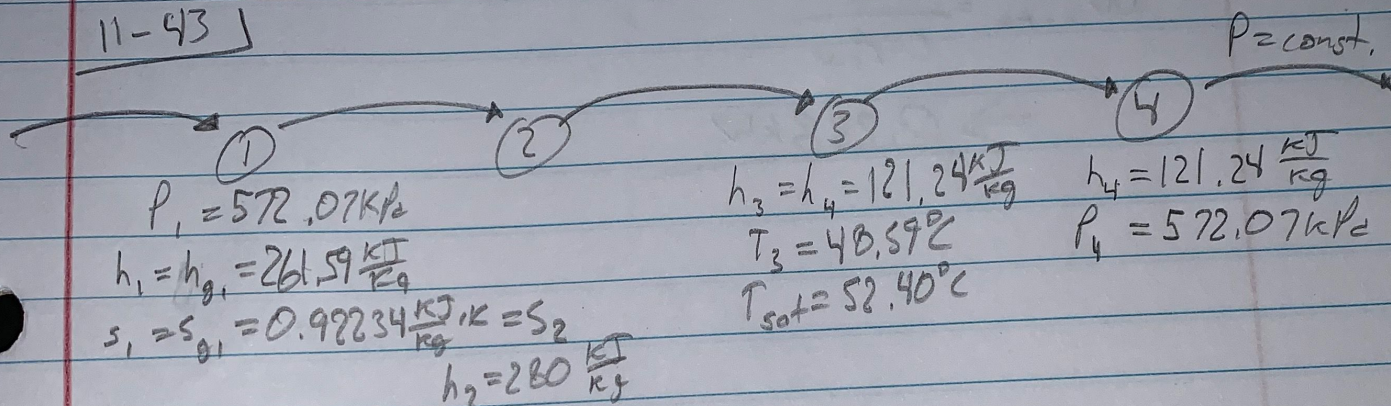
11-38)

From saturation tables for R-134a

$$P_{\text{sat}@-20^\circ\text{C}} = 132.82 \text{ kPa} \quad - \text{Recommended evaporator pressure}$$

$$P_{\text{sat}@35^\circ\text{C}} = 887.5 \text{ kPa} \quad - \text{Recommended condenser pressure}$$

11-43)



$$\Delta T_{\text{sub-cooling}} = T_{\text{sat}} - T_3$$

$$\Rightarrow 3.81^\circ\text{C}$$

$$\dot{W}_m = \dot{m}_R (h_2 - h_1) + \dot{Q}_{\text{loss}}$$

$$\Rightarrow 0.6564 \text{ kW}$$

$$h_{6E} = 209.34 \frac{\text{kJ}}{\text{kg}}$$

$$h_{6O} = 167.53 \frac{\text{kJ}}{\text{kg}}$$

$$\dot{Q}_H = \dot{m}_R (h_2 - h_3)$$

$$\Rightarrow 3.074 \text{ kW}$$

$$\dot{Q} = \dot{m} (h_{6E} - h_{6O})$$

$$\Rightarrow 2.178 \text{ kW}$$

$$\text{COP} = \frac{\dot{Q}_H}{\dot{W}_m}$$

$$\Rightarrow 4.683$$

$$\dot{m}_R = \frac{\dot{Q}}{h_1 - h_4}$$

$$\Rightarrow 0.01936 \text{ kg/s}$$



11-43 cont. 1

$$COP_{rev} = \frac{T_H}{T_H - T_L}$$

$$\Rightarrow \frac{323K}{323K - 298K}$$

$$\Rightarrow COP_{rev} = 12.92$$

$$\dot{W}_{in-mn} = \frac{\dot{Q}_H}{COP_{rev}}$$

$$\Rightarrow \underline{0.238kW}$$