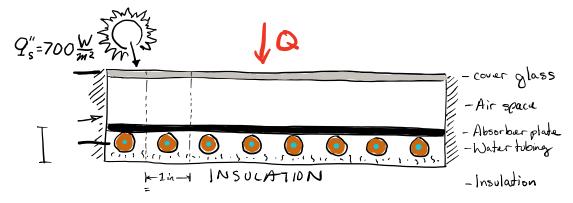
- 1. (55 points) A solar flux of 700 W/m2 is incident on a flat-plate solar collector used to heat water. 90% of the solar radiation passes through the cover glass and air, and is absorbed directly by the absorber plate. The remaining 10% is reflected away from the collector. Water flows through the tube passages on the backside of the absorber plate and is heated from an inlet temperature of 20oC. The convection coefficient of the water is 1000 W/m2.oK. The convection coefficient between the cover glass and the ambient air at 25oC is 50 W/m2.oK. The convection coefficient in the air space is 10 W/ m2.oK. The type K copper tubes are 1/2 in. nominal size (5/8 in. outside diameter and 0.528 in. inside diameter) and they are separated by 1 in. center-to-center. The tubes are buried in a 1 in. thick slab made of silver. The back of this silver slab is well insulated. The cover glass thickness is 1/8 in. Assume the absorber plate to be very thin and highly conductive. The air space is 2 in. thick. Doesn't offer much resistance to the circuit
  - A. What is the amount of heat collected by the water in one of the tubes?
  - B. What is the air space temperature?
  - C. If the outlet temperature of the water is 45oC, what is the flow rate? Assume the specific heat of the water to be 4179 J/kg.oK.
  - D. The collector efficiency  $\eta$  is defined as the ratio of the useful heat collected to the rate at which solar energy is incident on the collector. What is the value of n?

```
PURPOSE:
```

Find out how much heat goes into the water flowing in the pipes

DRAWINGS & DIAGRAMS:



SOURCES:

-> Bayazitoglu, Y., Ozisik, N., "A textbook for heart transfer fundamentals", Begell House, Inc. (2012)

DESIGN CONSIDERATIONS:  
(assumptions, safety, cost, etc.)  

$$A = (\lim)(Im)(Im) = (0.0254m)(Im) = 1000 \text{ w}[(m^2 \cdot k)] = 1000 \text{$$

DATA & VARIABLES: -> See Design Considerations

PROCEDURE:  
1-A) What is the amount of heat collected by the water in one of the types?  
- Silver is the storting point. This problem resembles the iglue problem given in  
class where heat transfers in two different directions.  
- The first thing that actually absorbs any heat is the silver, so I will  
stort there. From there it goes up into the atmosphere, traveling through  
the absorbur glate, the eir pocket, and the cover glass.  
- Two shope factor eques will be used to solve, since heat transfers in two  
directions I = convertion 
$$\frac{L}{KA} = conduction$$
  
Toker  $\frac{1}{NA} = convertion \frac{2}{KA} = conduction$   
Toker  $\frac{1}{NA} = convertion \frac{1}{KA} = conduction \frac{1}{NA} = conduction$   
Toker  $\frac{1}{NA} = convertion \frac{1}{KA} = conduction \frac{1}{NA} = conduction$   
Toker  $\frac{1}{NA} = convertion \frac{1}{KA} = conduction \frac{1}{NA} = conduction \frac{1}{NA} = convertion \frac{1}{KA} = conduction \frac{1}{NA} = convertion \frac{1}{KA} = conduction \frac{1}{NA} = convertion \frac{1}{KA} = conduction \frac{1}{NA} = convertion \frac{1}{NA} = convertion \frac{1}{KA} = conduction \frac{1}{NA} = convertion \frac{1}{NA} = convertion \frac{1}{KA} = convertion \frac{1}{$ 

this is Lglass in metric

$$f_{eq_{1}} = \frac{1}{(100)[(m^{2}.^{\circ}K))(0.0254m^{2})} + \frac{1}{(100)[(m^{1}.^{\circ}K))(0.0254m^{2})} + \frac{0.003175m}{(0.780)[(m^{-}K))(0.0254m^{2})} + \frac{1}{(500)[(m^{2}.^{\circ}K))(0.0254m^{2})}$$

$$= 3.937 + 3.937 + 0.1603 + 0.7874$$

$$F_{eq_{1}} = 8.82 \text{ k/W}$$

$$Q_{1} = \frac{AT}{Req_{1}} = \text{We'll put Req, where it belongs a little bit later}$$

-> Now, we solve for keq2. In order to do that, we must solve for the shape factors that are unknown in the keq2 equation.

$$\int z = \frac{1}{12} \int z$$

$$F = (5|8in) | 2 = 0.3125in = .0079375M$$

$$F = 0.5in = .0127M$$

$$S_{1} = \frac{4\pi r}{1 + r/(2z)} = \frac{4\pi (0.3125in)}{1 + (0.3125in)/(2(0/5in))}$$

$$S_{1} = \frac{4\pi r}{1 + r/(2z)} = \frac{4\pi (.0079375m)}{1 + (.0079375m)} = 0.076m$$

$$S_{1} = 0.076m$$

$$S_{1} = 0.076m$$

Shape factor problem of copper-to water (case #2)  

$$S_{2} = \frac{2 \pi H}{\ln (r_{2}/r_{1})}$$

$$S_{2} = \frac{2 \pi (1m)}{\ln (c_{0}H3 Rsm/ 0.067054m)}$$

$$F_{2} = (S|Bin)/2$$

$$= 0.2644 in$$

$$= 0.20647056m$$

$$F_{2} = (S|Bin)/2$$

$$= 0.3125in$$

$$= 0.0071375m$$

$$S_{2} = 37.25m$$

$$\Rightarrow Now we can solve for Reals$$

$$F_{el_{2}} = \frac{1}{S_{2}k_{starr}} + \frac{1}{S_{2}k_{starr}} + \frac{1}{h_{a_{1}0}A_{starr}}$$

$$F_{el_{2}} = \frac{1}{(countyr)(u^{th}w)(v^{t}x)} + (S^{T}, S^{T})(S^{T})(s^{T}) + \frac{1}{(countyr)(u^{t}, x)}$$

$$F_{el_{2}} = 0.353885 k/M$$

$$h_{utr} = 100 k/2^{2} k$$

$$h_{utr} = 10 k/2^{2} k$$

$$h_{utr} = 100 k/2^{2} k$$

$$h_{utr} = 10 k/2^{2} k^{2} k$$

$$h_{utr} = 100 k/2^{2} k^{2} k$$

$$h_{utr} = 100 k/2^{2} k^{2} k$$

$$h_{utr} = 10 k/2^{2} k^{2} k^{2} k$$

$$h_{utr} = 10 k/2^{2} k^{2} k^{2} k$$

$$h_{utr} = 10 k/2^{2} k^{2} k^{2} k^{2} k^{2} k^{2} k^{2} k^{2} k^{2} k^{$$

$$Q_{1} = \frac{\Delta T}{Raq_{1}} = \frac{T_{\text{silver}} - T_{\infty}}{\frac{1}{hA_{\text{silver}}} + \frac{1}{hA_{\text{glass}}} + \frac{L}{kA_{\text{glass}}} + \frac{1}{hA_{\text{glass}}}} + (+) = Q_{T}$$

$$Q_{2} = \frac{\Delta T}{Raq_{2}} = \frac{T_{\text{silver}} - T_{\text{water}}}{\frac{L}{kA_{\text{silver}}} + \frac{L}{hA_{\text{soperr}}}} + \frac{1}{hA_{\text{soperr}}} + \frac{1}{hA_{\text{soperr}}}}$$
These two have been changed with the shape factor solutions

$$Q_{1} = \frac{\Lambda T}{Req_{1}} = \frac{1}{hA_{silver}} + \frac{1}{hA_{glass}} + \frac{L}{kA_{glass}} + \frac{1}{hA_{glass}} + \frac{1$$

$$Q_7 = (700W)(0.90) = 630W = Q_1 + Q_2 = \frac{T_{silver} - 298K}{8.82KW} + \frac{T_{silver} - 293K}{0.353885KW}$$
  
 $Q_7 = 630W = \frac{T_{silver} - 298}{8.82WW} + \frac{T_{silver} - 293K}{0.353885KW}$ 

$$\underbrace{ \begin{array}{c} 0.353885 \\ \hline 0.353885 \end{array} } \underbrace{ \left( \frac{T_{silver} - 218}{8.82} \right) + \underbrace{ \left( \frac{8.82}{8.82} \right) \underbrace{ \left( \frac{T_{silver} - 293}{8.82} \right) }_{0.353885} }$$

$$G_{30} = \frac{0.3538857_{5} - 105.458}{3.12127} + \frac{8.827_{5} - 2584.24}{3.12127}$$

$$630 = 9.17389T_{s} - 2689.72$$
  
 $3.12127$ 

$$I491.74 = 2.93915 T_{s}$$

$$T_{s} = 507.541K = 234.391°C = This is the silver temperature due to the total heat absorption
$$Q_{1} = \frac{T_{s}ilver - T_{oo}}{R_{eq}} = \frac{234.391°C - 25°C}{8.82°K/W}$$

$$Q_{1} = 23.7405W$$

$$Q_{2} = \frac{T_{s}ilver - T_{H_{2}O}}{R_{eq_{2}}} = \frac{234.391°C - 20°C}{0.353885°K/W} = 605.821W$$

$$Q_{2} = 605.821W$$

$$ANSWER$$$$

$$I-B$$
) what is the air space temperature?  
 $Q = \frac{\Delta T}{R}$ 

$$\frac{T_{\text{silves}} - T_{\text{air} \text{pocket}}}{K_{\text{silverbair}}^{\text{con}}} = \frac{234.391^{\circ}\text{c} - T_{\text{air}}}{\left(\frac{1}{(10w \left[ lw^{2.\circ} k \right])(0.0254 \mu r^{2})} \right)} = 23.7405W$$

$$T_{\text{air}} = 234.391 - (23.7405)(3.9370)$$

$$T_{\text{air}} = 140.924^{\circ}\text{c}$$
ANSWER

1-c) If the outlet temperature of the water is 45°C, what is the flow rate? Assume the specific heat of the water to be 4179 J/kg. K

$$\begin{bmatrix} Q_{H_{2}0} = \dot{m} (v_{H_{2}b} \Delta T) \\ 45^{\circ} (-70^{\circ} (-70^{\circ}$$

1-D) The collector efficiency (n) is defined a	as the ratio of the useful heart collected
to the rate at which solar energy is	is incident on the collector. What is the
value of n?	before the percentage of absorption
$\mathcal{M} = \frac{Q_{H_{20}}}{Q_{1}} = \frac{605.821W}{700W}$	

Y = 0.87 ANSWER

ANALYSIS:

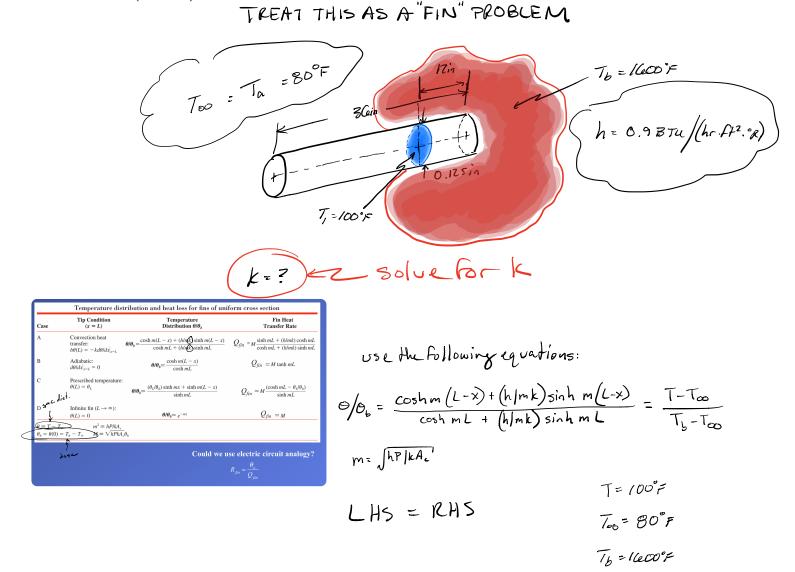
1-A) Units cancelled nicely, and the correct units of W is in place. This is the amount of heat that was transferred from the ambient air outside the collector into the water flowing in the pipes.

1-B) This value is the temperature of the air pocket between the collector and the glass. When it's coming in, the heat passes through the air and collector and is essentially fully absorbed by the silver. From there (thesilver), heat overflows back into the air space between the collector and the glass

1-() This is the mass flow rate of the water flowing through the pipes embedded in the silver. Given the know ledge of the specific heat of the fluid, and what temperature it heats to, we can pair it with the calculated heat (in wats) transferred to it in 1-A and then solve for the mass flow rate.

1-D) Efficiency is always unitless, as is my answer.

2. (25 points) The brazing operation is a metal-joining process that is done by heating a base metal to a high temperature and applying a brazing material to the heated joint. The heated base metal melts the brazing alloy, which fills the joint, and then solidifies when allowed to cool. The brazing material is a copper alloy in the shape of a rod of 1/8 in. diameter and 3 ft length. The end of the brazing rod (where it is being melted) reaches an average temperature of 1600 oF. What is the conductivity of the brazing material if at a distance of 1 ft the temperature is 100 oF? The ambient temperature is 80 oF, and the convection coefficient is 0.9 BTU/(hr.ft2.oR).



 $\frac{100 - 80}{1600 - 80} = \frac{20}{1520} = 0.0132$ 

$$\frac{\Theta}{\Theta_b} = \frac{T - T_{\infty}}{T_b - T_{\infty}} = \frac{\cosh \int hP | kA_c' (L - x) + (h | (\int hP | kA_c' x k)) \sin h \int hP | kA_c' (L - x)}{\cosh \int hP | kA_c' L + (h | (\int hP | kA_c' x k)) \sinh \int hP | kA_c' L}$$

-> set up excel file to solve for k (see attached submission).

I tried to use Excel's Goal Seek function to find the value of k by setting the "RHS" equal to the "LHS". The closest value of k that was computed was  $k = \frac{6}{20} BTU / (h \cdot ft \cdot F)$