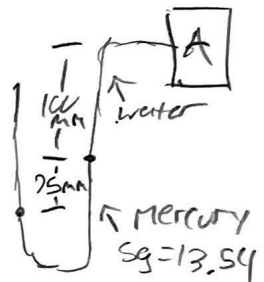


11.) $\Delta P = \gamma h$ $\gamma = 0.076 \text{ lb/ft}^3$ $h = 4000 \text{ ft}$ $P = 14.7 \text{ atm}$
 $\Delta P = (0.076)(4000) = 304 \text{ lb/ft}^2 \times (\text{lb/ft}^2 / 12 \text{ in})$
 $\Delta P = 2.11 \text{ psi} = 14.7 \text{ atm}$
 $P = 12.59 \text{ psi}$

13.) Ogauge, Since it is at 1 atm pressure at the surface

41.) $P_{\text{air}} + (\gamma h_o) = 180 \text{ psig}$ $\gamma = 0.9$
 $P_{\text{air}} + (0.90) \times (32 + (80 - 48)) = 180 \text{ psig}$
 $P_{\text{air}} = 180 - 57.6$
 $P_{\text{air}} = 122.4 \text{ psi}$

62.) $P_2 = P_A + \rho_w g h_1 + \rho_m g h_2$
 $= P_A + (1000 \times 9.81 \times 0.1 \text{ m}) + (13600 \times 9.81 \times 0.075 \text{ m})$
 $= P_A + (981) + (10,006.2)$
 $\Rightarrow P_2 = P_A + 10,987.2 \Rightarrow P_A = 10,987.2 \text{ Pa}$



- 6.) True
 7.) False, atmospheric pressure will also change
 8.) False, the absolute pressure is always positive, perfect vacuum is 0
 9.) True
 10.) False, atmospheric pressure near surface is 101 kPa

83.) $P_{\text{atm}} = 30.65 \text{ in of Hg}$
 $P_{\text{atm}} = 30.65 \times 0.49$
 $P_{\text{atm}} = 15.018 \text{ psia}$

1 in of Hg = 0.49 psia

90.) $P = -68.2 \text{ kPa}$
 $68.2 \text{ kPa} \times 1000 = 6820 \text{ Pa}$
 $\frac{6820 \text{ Pa}}{133.322} = 51.1543 \text{ mm Hg}$

1 kPa = 1000 Pa
 $133.322 \text{ Pa} = 1 \text{ mm Hg}$

94.) $P_1 + \frac{1}{2} \rho V_1^2 + \rho g h_1 = P_2 + \frac{1}{2} \rho V_2^2 + \rho g h_2$
 $P_A + \rho g \times 16 \text{ m} = P_2$
 $P_2 = P_A + 1000 \times 9.81 \times 16$
 $= P_A + 156960 \text{ Pa}$

$P_{2 \text{ gauge}} = P_2 - P_A$
 $= 156960 \text{ Pa} \times 0.001$
 $= 156.96 \text{ kPa}$

$P_{\text{min}} = 156.96 \text{ kPa}$

