

Homework 1.2

MET 330

Due 09/12/19

Group 14

Zach Hollifield

Aaron Jackson

Nathanael Yapnayan

HW 1.2

Problem 1: From the solved problems in class, I learned that Pascal's paradox states that the pressure changes with elevation $\Delta P = \gamma h$. Pressure within a fluid is positive as you go down and is negative in the upward direction. It must be in the vertical direction. Pressure will be the same moving horizontally.

Pressure is the same at the bottom of all containers if the same fluid is in all the containers. If the density of a fluid is not given in the problem it can be found in the back of the book. When solving the problems, it does not matter which end you began at. Both ways will result in the same answers.

- 6) The value for the absolute pressure will always be greater than that for the gage pressure. True
- 7) As long as you stay on the surface of Earth, the atmospheric pressure will be 14.7 psia. False The magnitude of the atmospheric pressure varies with location and with climate conditions.
- 8) The pressure in a certain tank is -55.8 Pa (abs). False Absolute pressure is positive.
- 9) The pressure in a certain tank is -55.8 Pa (abs). False Absolute pressure is positive. Same at #8 on ebook?
- 10) The pressure in a certain tank is -150 kPa (gage). False Pressure -150 kPa (gage) is negative.

- 11) If you were to ride in a open-cockpit airplane to an elevation of 4000 ft above sea level, what would the atmosphere pressure be if it conforms to the standard atmosphere?

$$P_{\text{atmosphere}} = 14.7 \text{ psia} \quad h = \text{elevation} = 4000 \text{ ft}$$

$$\gamma_{\text{air}} = 0.0764 \text{ lb/ft}^3$$

$$P = \gamma h \rightarrow P = (0.0764 \text{ lb/ft}^3)(4000 \text{ ft})\left(\frac{1 \text{ lb}}{144 \text{ in}^2}\right) = 2.12 \text{ psia}$$

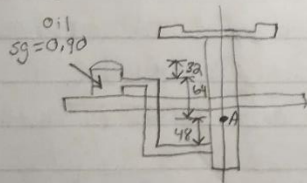
$$P = P_{\text{atm}} - \Delta P \rightarrow P = 14.7 \text{ psia} - 2.12 \text{ psia} = \boxed{12.6 \text{ psia}}$$

- 13) Expressed as gage pressure, what is the pressure at the surface of a glass of milk?

$$P_{\text{gage}} = P_{\text{abs}} - P_{\text{atm}} \quad P_{\text{abs}} = P_{\text{atm}} + \rho g(h) \rightarrow P_{\text{abs}} = P_{\text{atm}}$$

$$\text{so if } P_{\text{abs}} = P_{\text{atm}} \text{ then pressure at the surface } \boxed{= 0}$$

- 41) A diagram shows a hydraulic system for a vehicle lift. An air compressor maintains pressure above the oil in the reservoir. What must the air pressure be if the pressure at point A must be at least 180 psig?



$$\Delta P = \gamma_{\text{oil}} h$$

$$S_g = \frac{\gamma_{\text{oil}}}{\gamma_{\text{water}}}$$

$$h = 64 \text{ in}$$

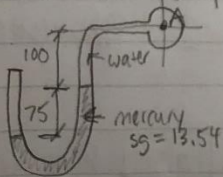
$$0.9 = \frac{\gamma_{\text{oil}}}{62.4 \text{ lb/ft}^3}$$

$$\gamma_{\text{oil}} = (0.90)(62.4 \text{ lb/ft}^3)\left(\frac{1 \text{ lb}}{144 \text{ in}^2}\right) = 0.033 \text{ lb/in}^2$$

$$\Delta P = (0.033 \text{ lb/in}^2)(64 \text{ in}) = 2.112 \text{ lb/in}^2$$

$$P_{\text{air}} = 180 \text{ psig} - 2.112 \text{ psig} = \boxed{177.9 \text{ psig}}$$

- 62) Calculate the pressure at point A in kPa (gage)



$$P_1 - 0.075 \text{ m } \gamma_{\text{Hg}} - 0.1 \text{ m } \gamma_{\text{H}_2\text{O}} = P_A$$

$$\gamma_{\text{H}_2\text{O}} = 9.81 \text{ kN/m}^3$$

$$0 - 0.075 \text{ m } (13.54) - 0.1 \text{ m } (9.81) = P_A$$

$$\boxed{P_A = -1.99 \text{ kPa}}$$

- 83) A barometer indicates the atmosphere pressure to be 30.65 in of mercury. Calculate the atmospheric pressure in psia. Pressure 1 inch of mercury = 0.491154 psia
$$P_{atm} = 30.65 \text{ in Hg} \times \left(\frac{0.491154 \text{ psia}}{1 \text{ in of Hg}} \right) = \boxed{15.05 \text{ psia}}$$

- 90) The pressure in a vacuum chamber is -68.2 kPa. Express this pressure in mmHg. $-68.2 \text{ kPa} = -68200 \text{ Pa}$
 $1 \text{ Pa} = 0.0075 \text{ mm of mercury}$ $-68200 \times 0.0075 = \boxed{-511.5 \text{ mmHg}}$

- 94) A passive solar water heater is to be installed on the roof of a multi-story building. The heater tank is open to atmospheric pressure and is mounted 16m above ground level. In the static state what gage pressure in kPa must the plumbing line be designed to withstand if it is connected all the way down to ground level?

$$\Delta P = \gamma h \quad h = 16\text{m} \quad \gamma_{\text{water}} = 9.81 \text{ kN/m}^3$$

$$\Delta P = (9.81 \text{ kN/m}^3)(16\text{m}) = \boxed{156.96 \text{ kPa}}$$

HW 1,2

1 Write a paragraph or two on what you learned

2 Chapter 3: 6-10, 11, 13, 41, 62, 83, 90, 94

1 We learned about pascals paradox $\Delta P = \gamma h$, otherwise known as the "gamma h" equation. We also learned about how pressure in a fluid is positive as you descend and negative as you ascend but is the same if you move horizontal. Density of fluids can be found in the back of the book if not given in the problem

3.6 True

3.7 False because atmospheric pressure changes with location or elevation

3.8 False because absolute pressure is positive

3.9 False because absolute pressure is positive

3.10 False negative gauge pressure

3.11 $P_{atm} = 14.7 \text{ psia}$ $\Delta h = 4000 \text{ ft}$
 $\gamma_{air} = 0.0764 \text{ lb/ft}^3$

$$\Delta p = \gamma h$$

$$0.0764 \text{ lb/ft}^3 \cdot 4000 \text{ ft} = 305.6 \text{ lb/ft}^2 \cdot \frac{1 \text{ ft}^2}{144 \text{ in}^2} = 2.12 \text{ psia}$$

$$P_{atm} - \Delta P = 14.7 \text{ psia} - 2.12 \text{ psia} =$$

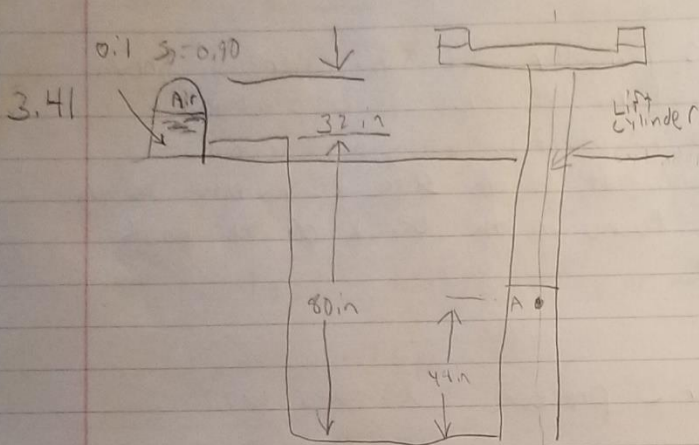
$$12.6 \text{ psia}$$

3.13 $P_{gauge} = P_{abs} - P_{atm}$

Open to atmosphere

$P_{abs} = P_{atm}$

Surface pressure = 0



$\Delta P = \gamma_{oil} h$ $s_{g, oil} = 0.90$

$s_g = \frac{\gamma_{oil}}{\gamma_{water}} \therefore 0.9 = \frac{\gamma_{oil}}{62.4 \text{ lb/ft}^3} = \gamma_{oil} = 56.2 \text{ lb/ft}^3 \cdot \frac{1}{12}$

$140 - 2.08 = 137.9 \text{ psi}$ $0.0325 \frac{\text{lb}}{\text{in}^3} \cdot 64 \text{ in} = 2.08 \text{ psi}$

$$3.62 \quad P_1 - 0.075 \text{ m } \gamma_{\text{ing}} - 0.1 \text{ m } \gamma_{\text{H}_2\text{O}} = P_2$$

$$0 - 0.075 \text{ m} \cdot 113.54 - 0.1 \cdot 9.81 = -1.99 \text{ kPa}$$

$$3.63 \quad 1 \text{ in Hg} = 0.49 \text{ psia} \quad 1 \text{ in Hg} = 30.65 \text{ in}$$

$$P_{\text{atm}} = 50.65 \text{ in Hg} \quad \frac{0.49}{1 \text{ in of Hg}} = 15.02 \text{ psia}$$

$$3.90 \quad 1 \text{ Pa} = 0.0075$$

$$P = -68.2 \text{ kPa}$$

$$-68.2 \cdot 0.0075 = -0.5115 \cdot 1000 = -511.5 \text{ mmHg}$$

$$3.94 \quad h = 16 \text{ m} \quad \gamma_w = 9.81 \text{ kN/m}^3$$

$$\Delta P = \gamma h$$

$$P = 9.81$$

$$P = 9.81 \text{ kN/m}^3 \cdot 16 \text{ m} = 156.96 \text{ kPa}$$

Hydrodynamics
Yagnason

1/2

Part 1

Pressure in a fluid increases as we go down.
Pressure decreases as we go up. It's like
going in on a swimming pool. Atmospheric
pressure is always going to be 0 kPa. When
we're discussing about the pressure - elevation
equation ($\Delta p = \gamma h$), the gamma (γ) can change
based on the elevation like moving into the atmosphere.
Pressure changes in a vertical plane. Horizontal
never changes in pressure.

During Tuesday's lecture, we learned more about
Hydrodynamics. We began to discuss about
streamlines. Streamlines are a path traced by a
massless particle moving with the flow which are
fluid particles. Velocity is tangent to streamline
point. Conservation of mass law was discussed.
The mass flow rate is the rate of mass passing
through an area. Velocity will always be
perpendicular to area. In conservation of mass
law, the change in mass equals the rate of change
over time. $(m_{in} - m_{out}) = \frac{dm}{dt}$

Problem 90

The pressure in a vacuum chamber is -68.2 kPa . Express this pressure in mm Hg.

Given

$$P = -68.2 \text{ kPa}$$

$$\text{Hg of } 1 \text{ Pa} = 0.0075 \text{ mm}$$

$$= -68.2 \text{ kPa} \left(\frac{1000}{\text{kPa}} \right) (0.0075 \text{ mm})$$

$$= \boxed{-511.5 \text{ mm Hg}}$$

Vacuum chamber

Problem 94

A passive solar water heater is to be installed on the roof of a multi-story building. The heater tank is open to atmospheric pressure and is mounted 16 m above ground level. In the static (non-flowing) state, what gage pressure, in kPa, must the plumbing line be designed to withstand if it is connected all the way down to ground level?

gage pressure - elevation equation

$$\Delta p = \gamma h$$

Given

$$\gamma_w = 9.81 \text{ kN/m}^3 \quad h = 16 \text{ m}$$

(water)

$$\Delta p = (9.81 \text{ kN/m}^3) (16 \text{ m})$$

$$\Delta p = 156.96 \text{ kN/m}^2$$

$$= \boxed{156.96 \text{ kPa}}$$

(gage)