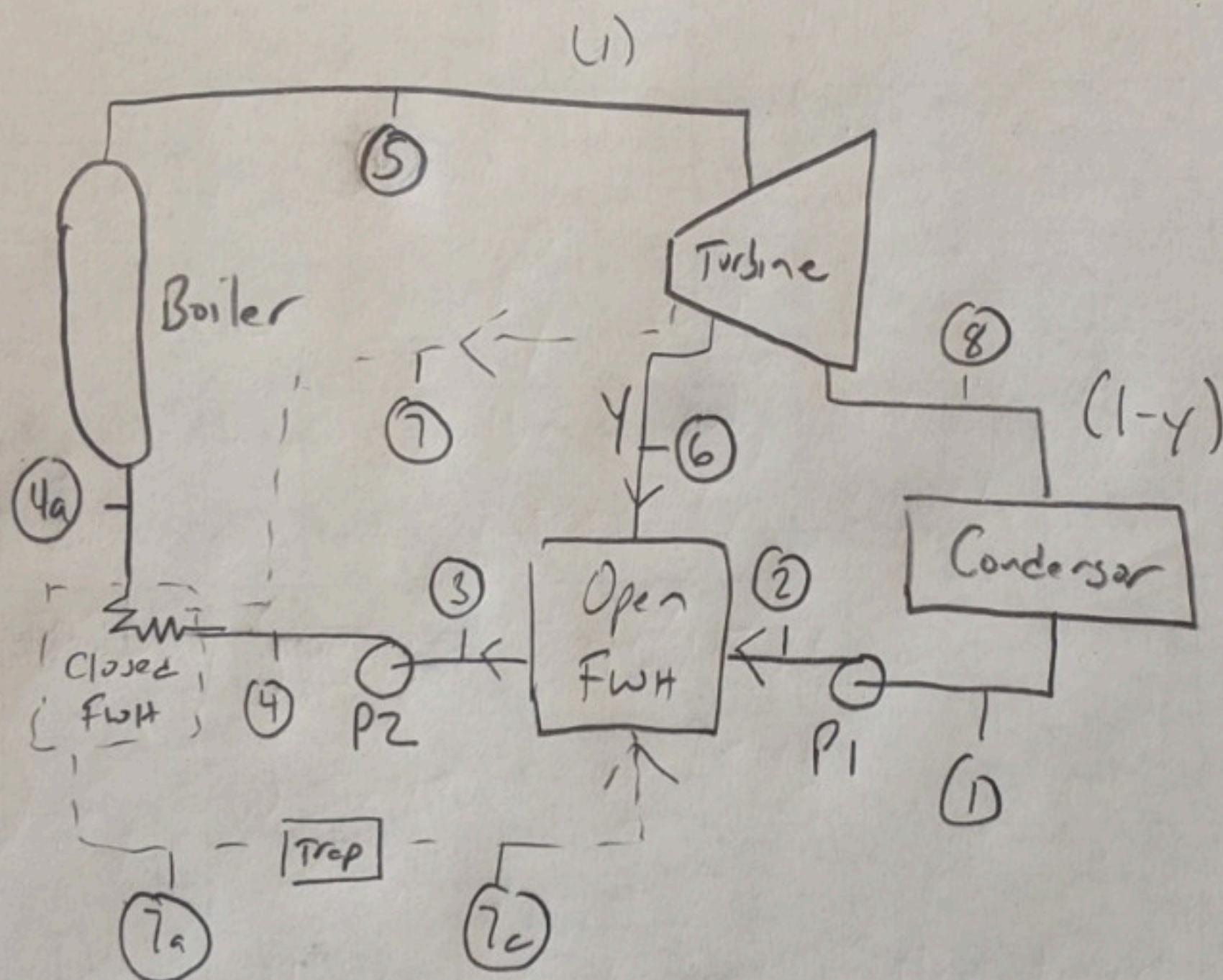


Purpose Calculate thermal efficiency and Specific work of an ideal steam regenerative cycle.

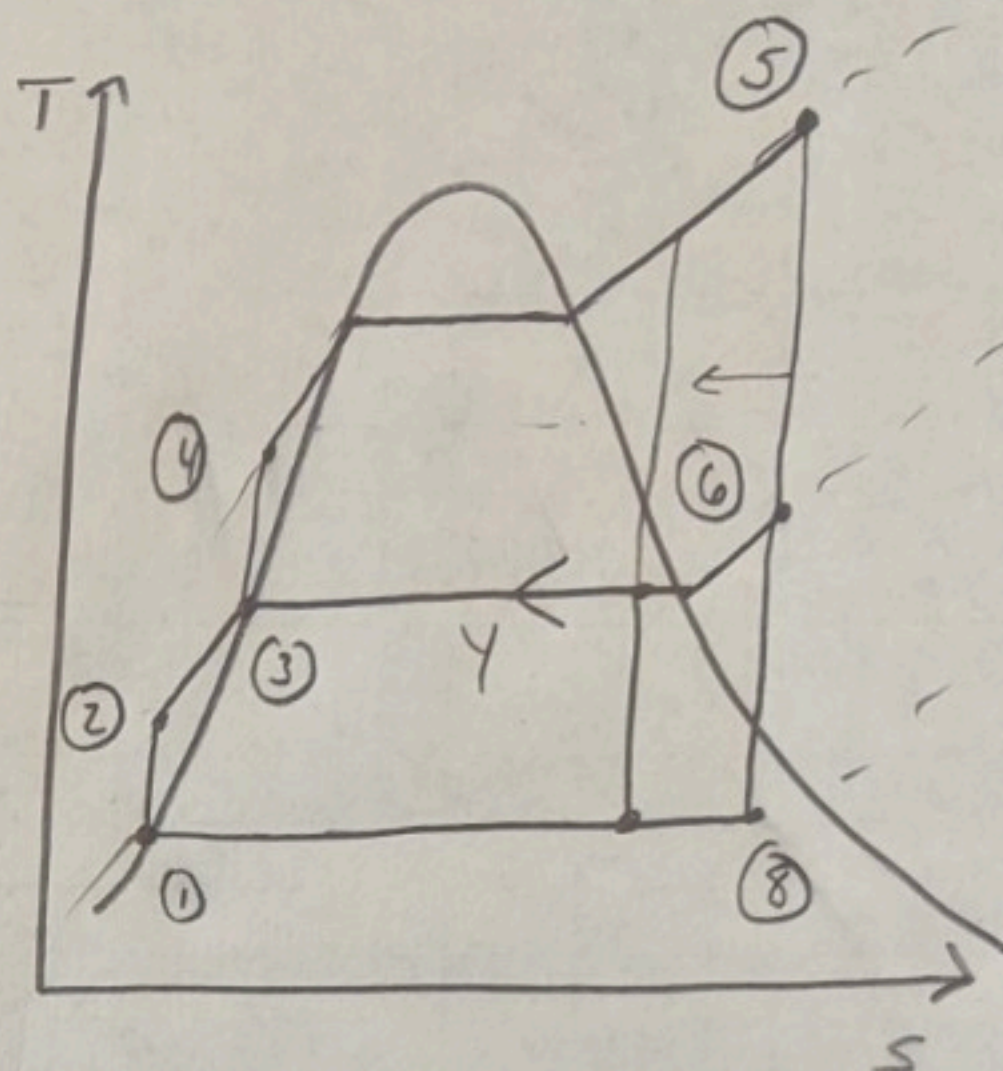
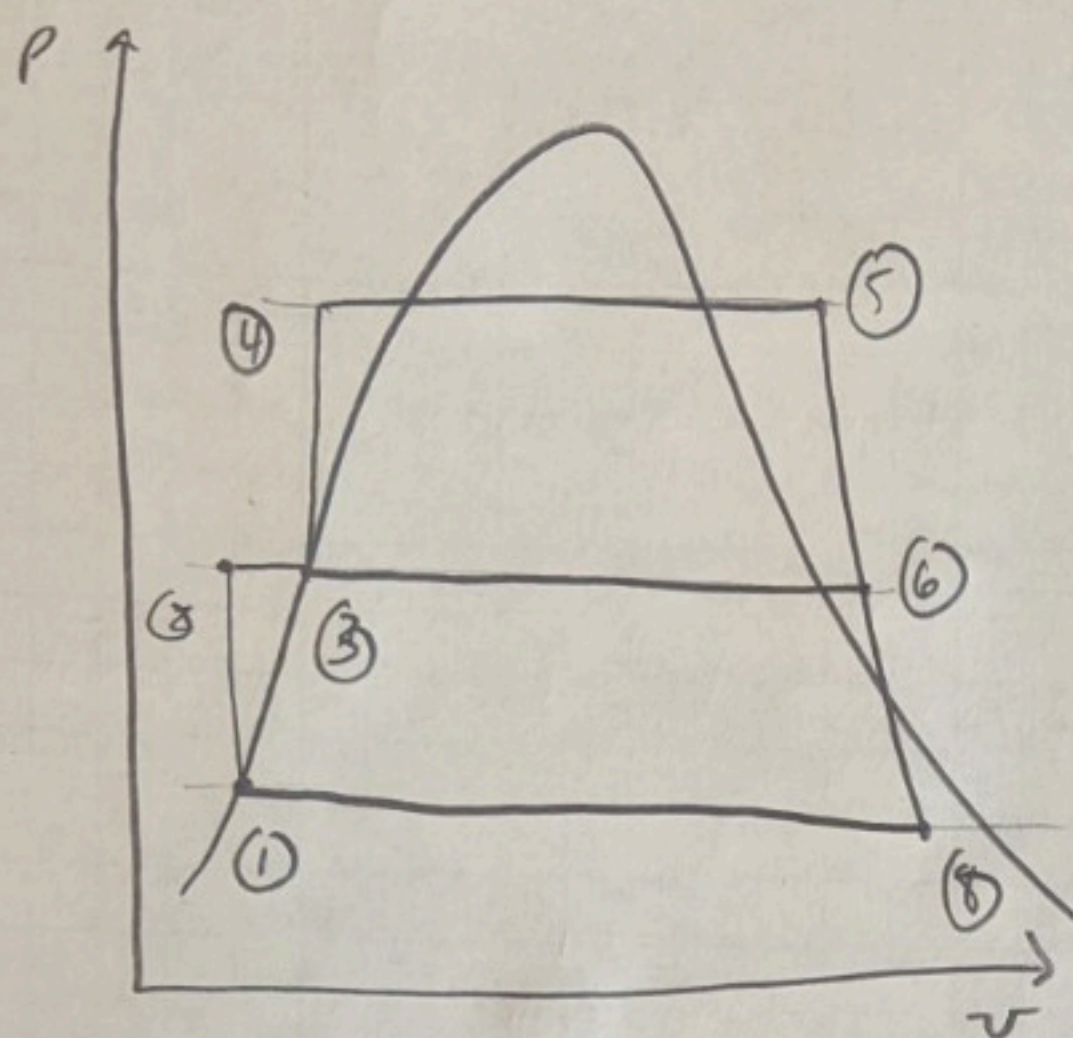
Drawings and diagrams



The steam trap is clogged, therefore no flow

through the pipe (7). $\dot{m} = 0 \Rightarrow \dot{Q}_{\text{Closed FWH}} = 0 = \dot{m}(\Delta h)$

$\Rightarrow$ Conditions at 4 = 4a, assuming no losses to ambient.



Sources

Cengel + Boles. "Thermodynamics - An Engineering Approach," 8th Edition, McGraw Hill, 2015

Design Considerations

- 1) Water is pure
- 2) Pumps and turbines are isentropic
- 3) Not heat losses in connections, pipes, neither fluid friction losses.

Data and Variables

Steam @ ⑤ = 3.0 MPa, 400°C

Steam @ ⑧ = 10 kPa

Steam @ ④ = 100 kPa

Water @ ①, ③, and ⑥ is saturated liquid. $\Rightarrow x=0$

Procedure

- ① Get states using Tables A-4, A-5, A-6
and the following equations

$$h_2 = h_1 + v_1 (P_2 - P_1) \quad \underline{\text{PUMPS ONLY}}$$

$$x_i = \frac{s_i - s_f}{s_{fg}}$$

$$h_i = h_f + x_i h_{fg}$$

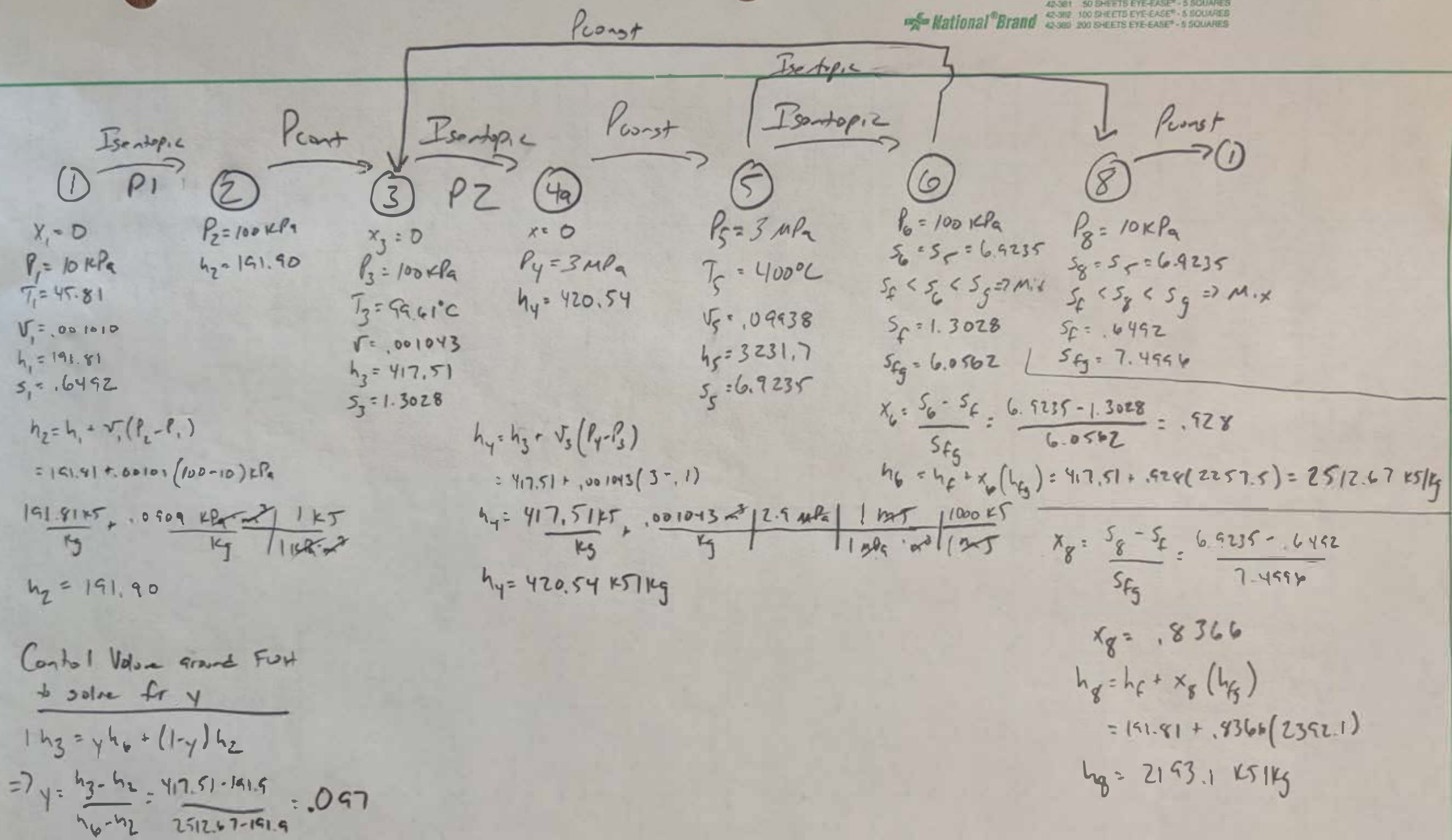
$$y = \frac{\dot{m}_6}{\dot{m}_5} = \frac{h_3 - h_2}{h_6 - h_2}$$

$$\dot{q}_{out} = (1 - y)(h_8 - h_1)$$

$$\dot{q}_{in} = h_5 - h_{4a}$$

$$\dot{w}_{net} = \dot{q}_{in} - \dot{q}_{out}$$

$$\eta_{th} = 1 - \frac{\dot{q}_{out}}{\dot{q}_{in}}$$



NET 350

TEST 3-1

ML SWERLUND

MET 350

TEST 3-1

ML SUNDERLAND

⑤

Calculate specific work and Thermal efficiency

$$\eta_f = \frac{w_{net}}{q_{in}}$$

$$= \frac{945.28}{2811.16}$$

$$\boxed{\eta_f = 33.63\%}$$

$$w_{net} = w_T - w_{p1} - w_{p2}$$

$$\begin{aligned} * w_T &= h_5 - y h_6 - (1-y) h_8 \\ &= 3231.7 - 0.97(2512.67) - 0.930(2193.1) \end{aligned}$$

$$w_T = 948.39 \text{ kJ/kg}$$

$$\begin{aligned} * w_{p1} &= (1-y) h_2 - h_1 \\ &= 0.930(191.9 - 191.81) \end{aligned}$$

$$w_{p1} = 0.0837 \text{ kJ/kg}$$

$$\begin{aligned} * w_{p2} &= 1(h_4 - h_3) \\ &= 420.54 - 417.51 \end{aligned}$$

$$w_{p2} = 3.03 \text{ kJ/kg}$$

$$* w_{net} = 948.39 - 0.0837 - 3.03$$

$$\boxed{w_{net} = 945.28 \text{ kJ/kg}}$$

$$\begin{aligned} q_{in} &= h_5 - h_4 \\ &= 3231.7 - 420.54 \end{aligned}$$

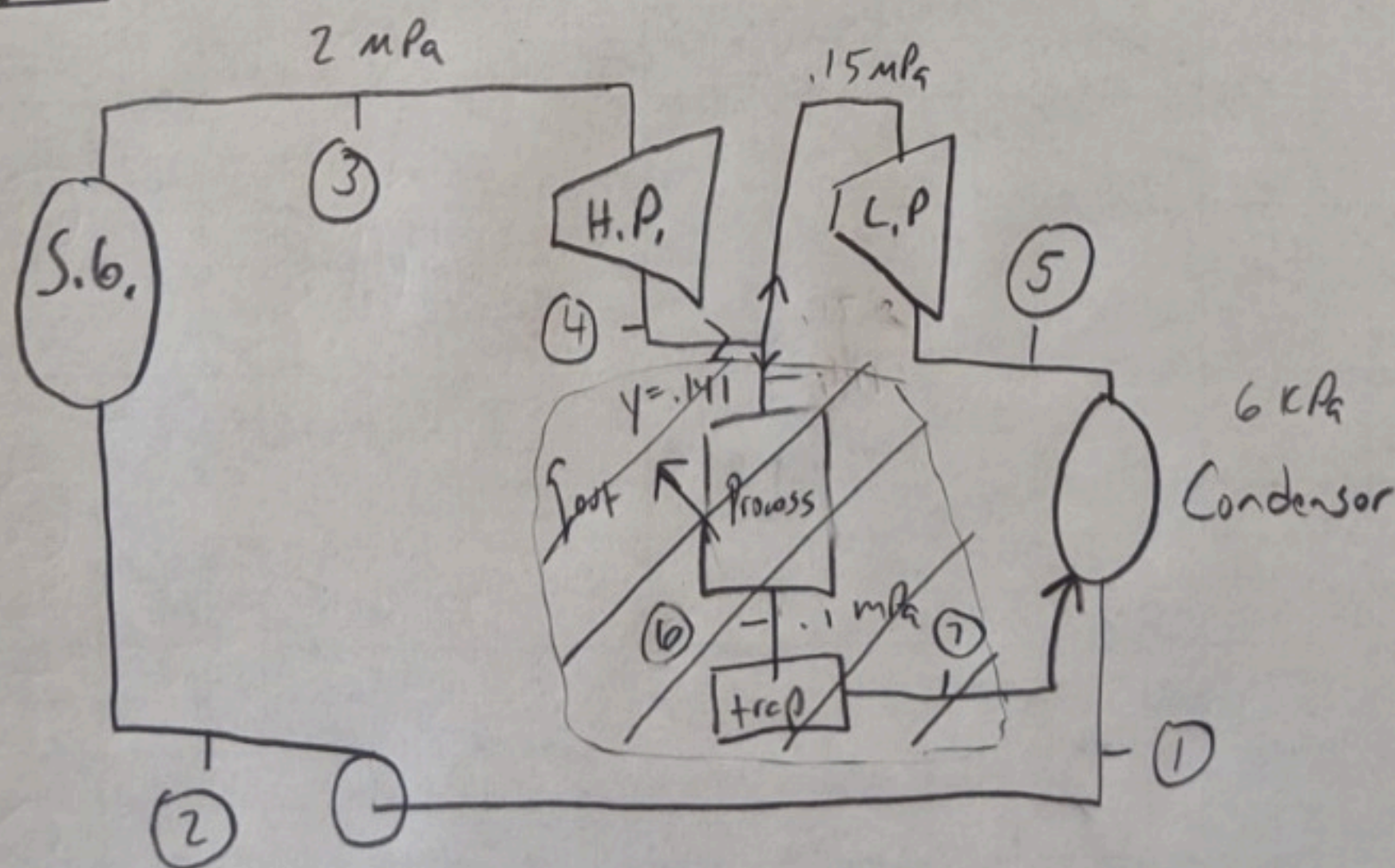
$$q_{in} = 2811.16 \text{ kJ/kg}$$

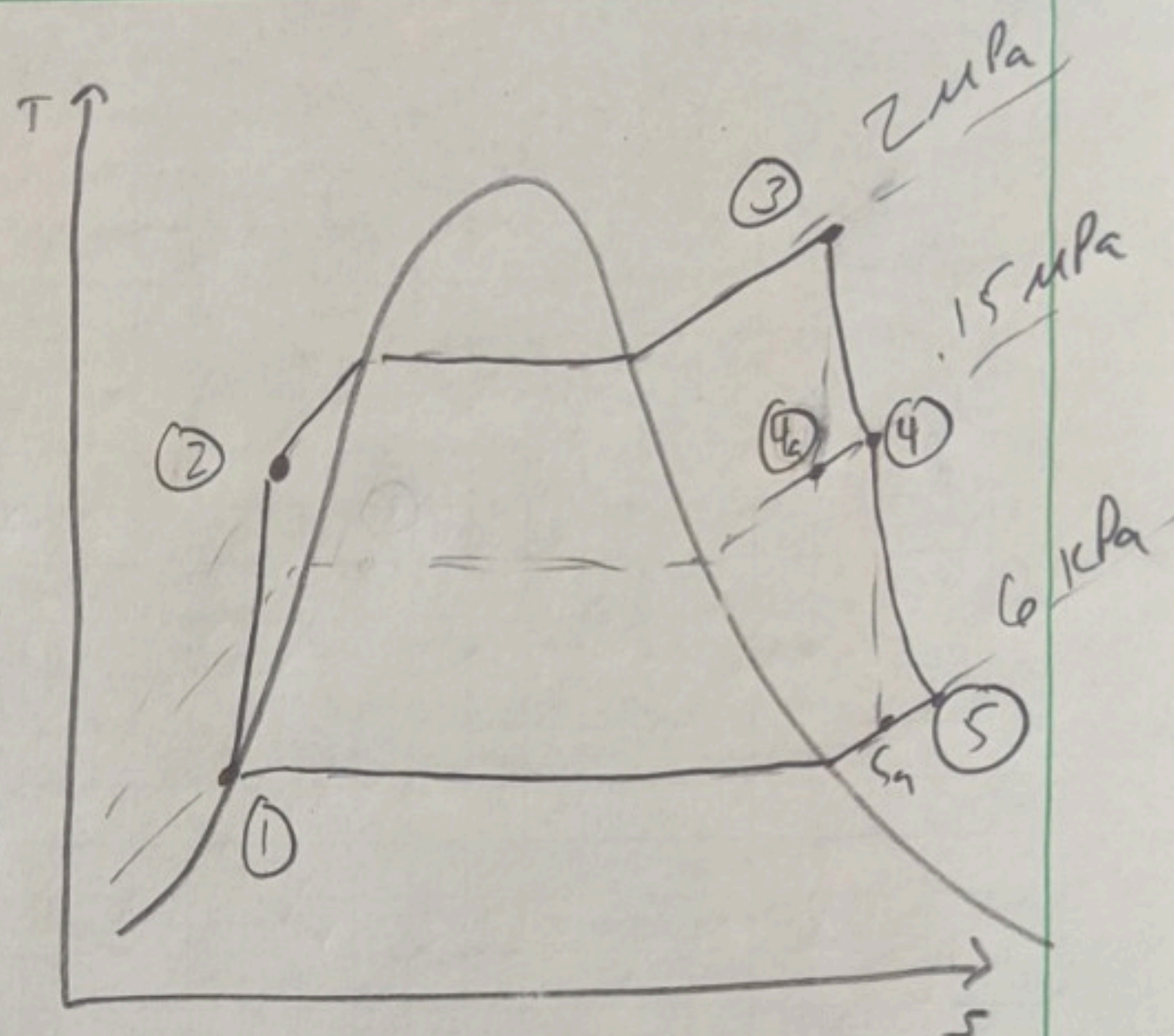
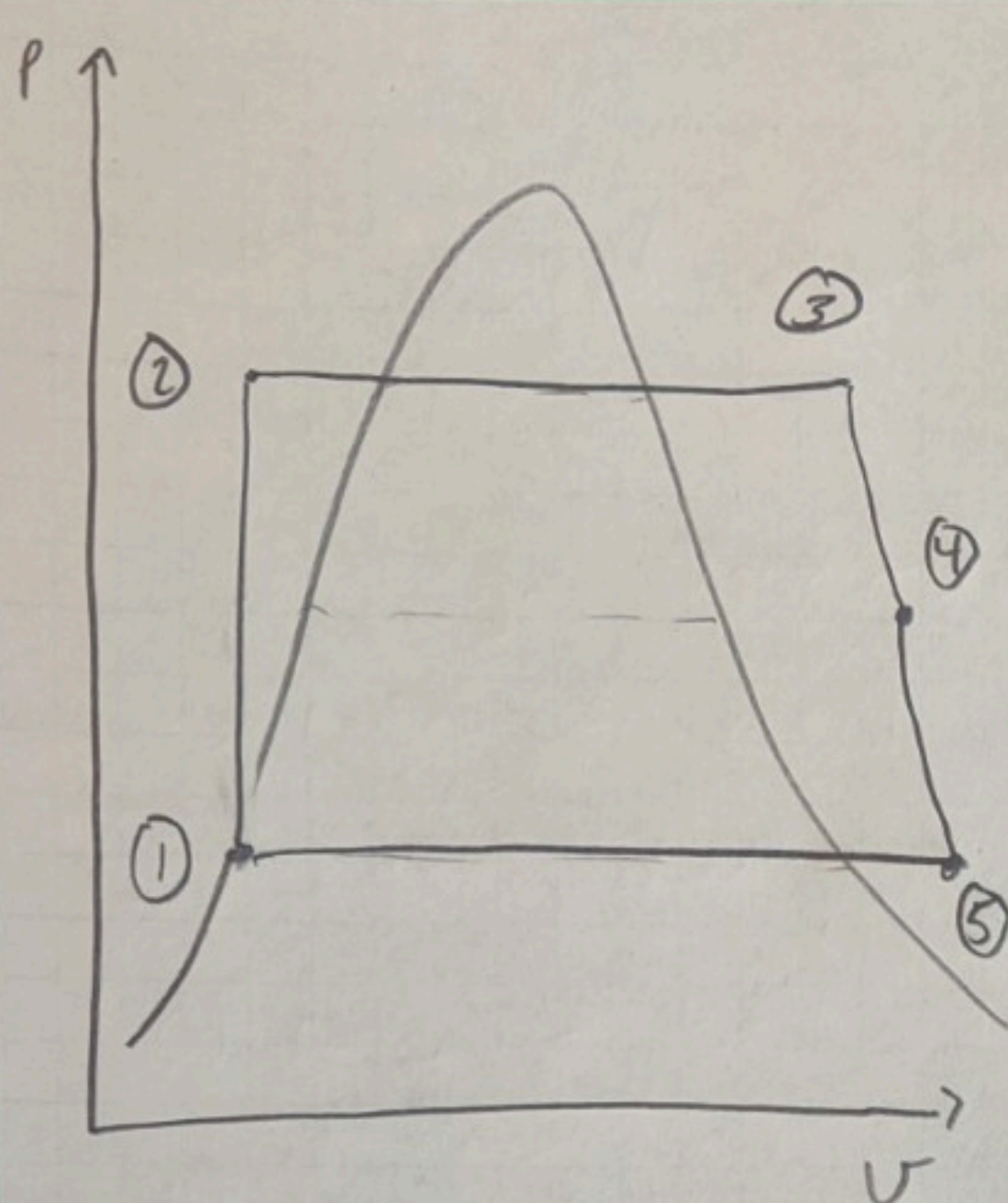
Purpose

Given specifications for a water
cogeneration steam cycle calculate the
following parameters if steam trap is closed!

- a) rate of heat transfer in Stm Gen.
- b) net power produced by turbines
- c) rate of heat transfer for building heating
- d) rate of heat transfer in condensor
- e) Utilization factor
- f) Thermal efficiency considering cycle does not heat building. (No stm extracted between HP + LP turbines.

Drawings





Because the steam trap is closed, there is no heat transfer in process and can be ignored in cycle diagrams

Source

Cengel + Boles, "Thermodynamics - An Engineering Approach", 8th Edition, McGraw-Hill, 2015

Design Considerations

- 1) Water is pure
- 2) Pump is isentropic
- 3) Turbines have $\eta_t = 0.8$
- 4) No heat losses in connectors or pipes. Disregard fluid friction losses

Data + Variables

Boiler produces steam @ 2 MPa, 320°C

$$\dot{m}_{\text{steam}} = 0.82 \text{ kg/s}$$

$$P @ \text{HP exhaust} = 0.15 \text{ MPa}$$

$$P @ \text{LP exhaust} = 6 \text{ kPa}$$

$$\eta_{\text{turbines}} = 80\%$$

Trip is blocked

Procedure

Get states Using Tables A-4, A-5, A-6

And the following equations

$$h_2 = h_1 + v(P_2 - P_1) \text{ Pump only}$$

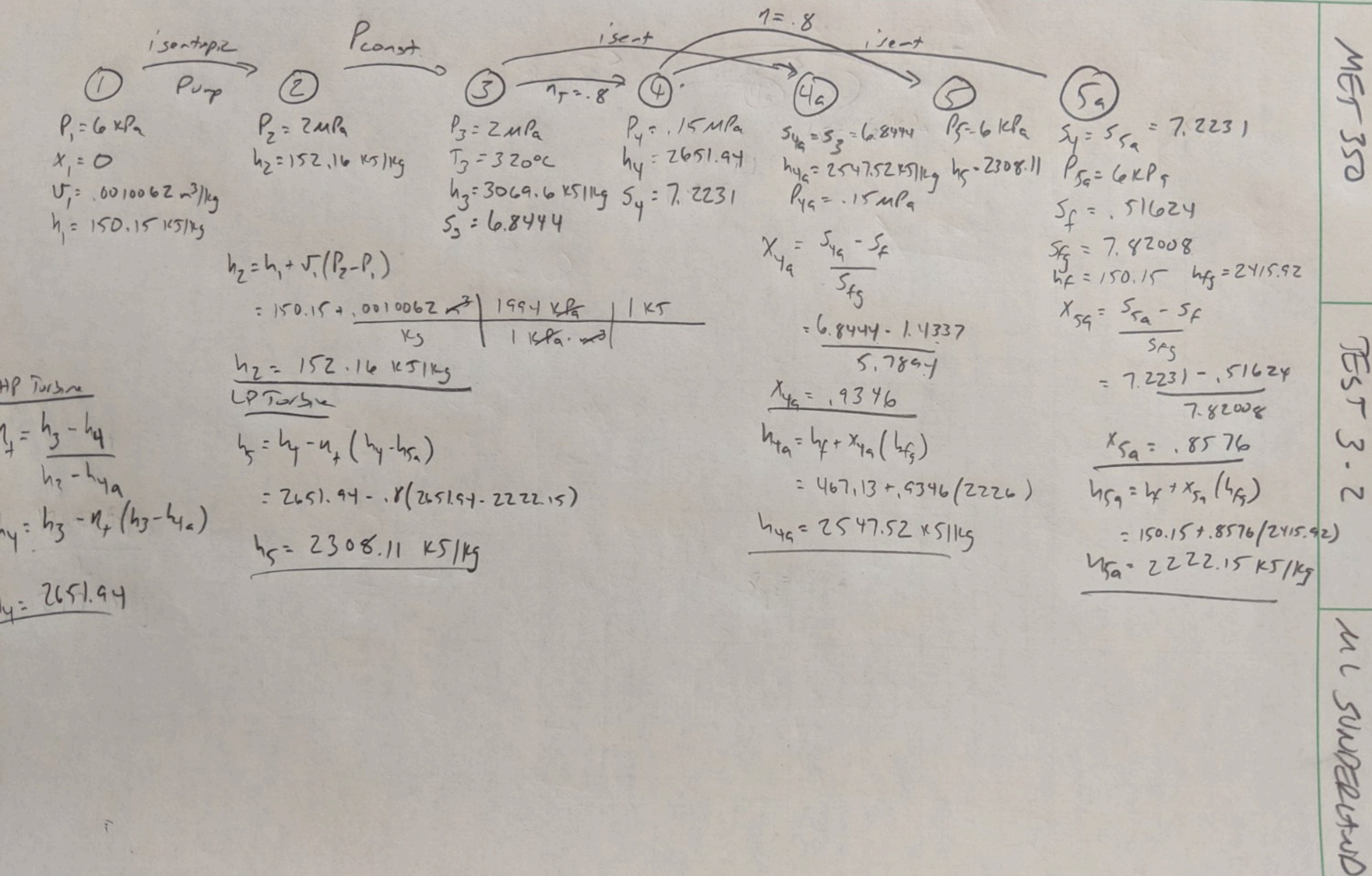
$$X_i = \frac{s_i - s_f}{s_{fg}} \quad h_i = h_f + X_i h_{fg}$$

$$\dot{q}_{\text{out}} = h_5 - h_1 = \dot{q}_{\text{cond}} \quad \dot{q}_{\text{in}} = h_3 - h_2 = \dot{q}_{\text{s.b.}}$$

$$E_u = \frac{\dot{W}_{\text{net}} + \dot{Q}_{\text{process}}}{\dot{Q}_{\text{in}}}$$

$$\dot{W} = \dot{m} g$$

$$\eta_t = \frac{h_3 - h_4}{h_3 - h_{4s}}$$



$$\dot{Q}_{s/g} = \dot{m}_{stm} (h_3 - h_2)$$

$$= \frac{.82 \text{ kg}}{\text{s}} \frac{(3069.6 - 152.16) \text{ kJ}}{\text{kg}}$$

a) $\dot{Q}_{s/g} = 2392.3 \text{ kW}$

$$\dot{W}_{net} = \dot{m} W_{net}$$

$$= \frac{.82 \text{ kg}}{\text{s}} \frac{759.48 \text{ kJ}}{\text{kg}}$$

b) $\dot{W}_{net} = 622.77 \text{ kW}$

c) $\dot{Q}_{process} = 0 \text{ kW}$

d) $\dot{Q}_{cond} = \dot{m} (h_5 - h_1)$

$$= \frac{.82 \text{ kg}}{\text{s}} \frac{(2308.11 - 150.15) \text{ kJ}}{\text{kg}}$$

$\dot{Q}_{cond} = 1769.53 \text{ kW}$

e) $\epsilon_u = \frac{\dot{W}_{net} + \dot{Q}_{process}}{\dot{Q}_{in}} = \frac{622.77}{2392.3} = .26 \Rightarrow \boxed{26\%}$

f) $\eta_t = \frac{\dot{W}_{net}}{\dot{Q}_{in}} = \boxed{26\%}$

$$W_{net} = W_{out} - W_{in}$$

$$W_{net} = 761.49 - 201 = 759.48 \text{ kJ/kg}$$

$$W_{out} = W_{T1} + W_2$$

$$= h_3 - h_4 + h_4 - h_5$$

$$= h_3 - h_5 = 3069.6 - 2308.11$$

$$W_{out} = 761.49 \text{ kJ/kg}$$

$$W_{in} = h_2 - h_1 = 152.16 - 150.15$$

$$W_{in} = 2.01 \text{ kJ/kg}$$