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Test 3 problem 1

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diagram

materials:

Atm. air @ 90°F

Water @ 55°F

air system @ 75°F

PURPOSE: quantity of air to CS (ma)
 * State of air finding "w" and "Tdb"
 * Draw on psychrometric chart
 * Capacity of dehumidifier "m_{wa}"

Data: given

$\dot{V}_{\text{exhaust}} = 500 \text{ ft}^3/\text{min}$

$\dot{V}_{\text{inlet}} = 500 \text{ ft}^3/\text{min}$

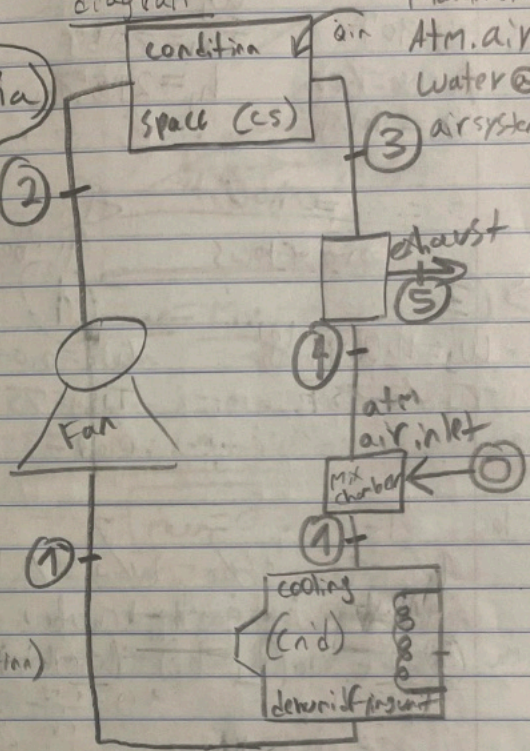
$T_{\text{db}} = 75^\circ\text{F}$ $T_{\text{air in}} = 90^\circ\text{F}$

$\phi_{\text{CS}} = 50\%$ $\phi_{\text{outside}} = 60\%$

$\dot{Q}_{\text{in}} = 72,000 \text{ Btu/hr}$

Assumptions:

- Constant mixing in (Cnd) unit
- humidity ratio in (CS) constant
- dehumidifying is ideal behavior
- fan is ideal (no Δ in air condition)



procedure:

1) States w/ Diagram

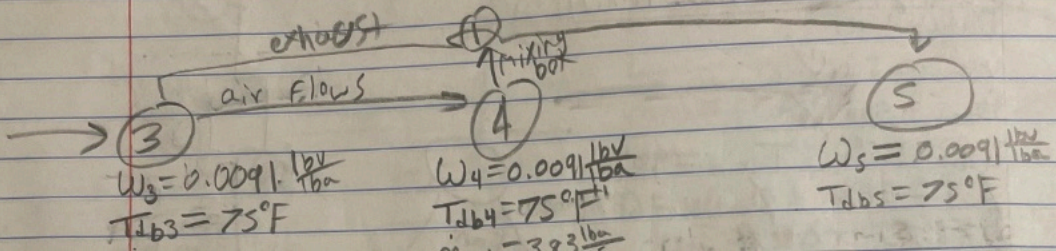
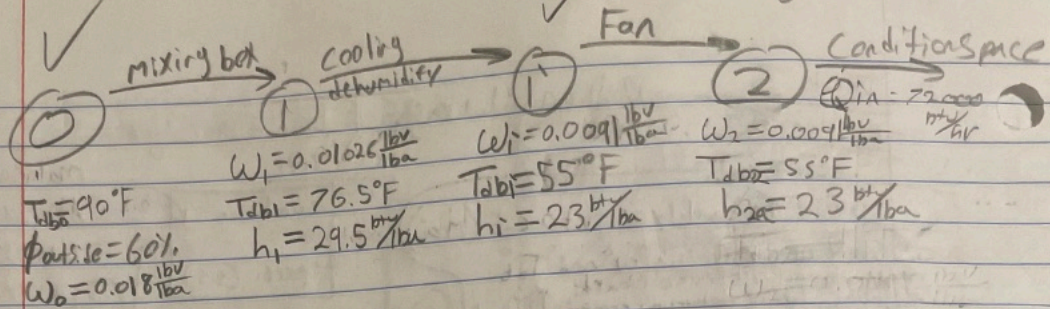
2) psychrometric chart $\rightarrow$ mark points, 2, 3, 4, 5, 1 $\rightarrow$ Find 1

3) calculations $\rightarrow$ 1st laws $\rightarrow$ conservation of mass $\rightarrow$ humidity ratio $\rightarrow$ 1st law

4) Solutions

Summary:

In problem 1, focusing on the conditioned space part of the air conditioning as a whole. as the cycle functions air (5) mixes w/ the air of the previous cycle (4) to combine and flow into a dehumidifier and cooling unit to the point where the dewpoint temp is reached and continues to cool into a fan and then into the confined space, having cold air enter to maintain a constant Temp. and relative humidity to balance the heat entering the room (conditioned space). after conditioned air leaves and heat is released outside into the atmosphere before accepting atmospheric air.



Calculations:

1st law

2-3

$\dot{m}_{a2} = \dot{m}_{a3}$

2nd law

1-1

$$\dot{Q}_{in} - \dot{Q}_{out} + \dot{m}_{in} \cdot h_{in} - \dot{m}_{out} \cdot h_{out} = \Delta h$$

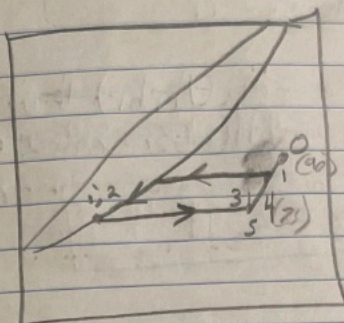
$$\dot{Q}_{in} = \dot{m}_{a3} \cdot h_3 - \dot{m}_{a2} \cdot h_2$$

dehumidification & cooling

$$\dot{Q}_{in} - \dot{Q}_{out} + \dot{m}_{in} \cdot h_{in} - \dot{m}_{out} \cdot h_{out} = \Delta h + h_w$$

$$\dot{Q}_{out} = \frac{\dot{m}_a (h_1 - h_2) - (\dot{m}_w \cdot h_w)}{\dot{m}_a}$$

$$\dot{Q}_{out} = \frac{h_1 - h_2}{\dot{m}_a} - (w_1 - w_2) \cdot h_w$$



$$\frac{\dot{m}_w}{\dot{m}_a} = \frac{\dot{m}_{w1} - \dot{m}_{w2}}{\dot{m}_a} \rightarrow w_1 - w_2$$

$$\dot{m}_w = \dot{m}_a (w_1 - w_2)$$

$$h_w \rightarrow \text{tables at } T_{db} = 55^\circ F$$

$$\dot{m}_w = 0.00515 \frac{\text{lb}_w}{\text{s}}$$

$$\dot{m}_{a3} = \dot{m}_{a4} + \dot{m}_{a5}$$

2-3 cont
psychrometric chart

$$\dot{Q}_{in} = \dot{m}_a (h_3 - h_2)$$

$$h_3 = 27.5 \frac{\text{Btu}}{\text{lb}_a} \quad h_2 = 23 \frac{\text{Btu}}{\text{lb}_a}$$

$$\frac{20 \frac{\text{Btu}}{\text{s}}}{4.5 \frac{\text{Btu}}{\text{lb}_a}} = \dot{m}_a = 4.44 \frac{\text{lb}_a}{\text{s}}$$

Mass Flow rates

$$\dot{m}_{a5} = \frac{1}{V_5} \cdot \dot{V} = \frac{1}{13.66 \frac{\text{ft}^3}{\text{lb}_a}} \cdot 8.33 \frac{\text{ft}^3}{\text{s}}$$

$$V_5 = 13.66 \quad \dot{V} = 8.33 \frac{\text{ft}^3}{\text{s}} \quad \dot{m}_{a5} = 0.61 \frac{\text{lb}_a}{\text{s}}$$

$$\dot{m}_{a1} = \dot{m}_{a3} + \dot{m}_{a4}$$

$$\dot{m}_{a4} = \dot{m}_{a3} - \dot{m}_{a5}$$

$$= 4.44 \frac{\text{lb}_a}{\text{s}} - 0.61 \frac{\text{lb}_a}{\text{s}}$$

$$\dot{m}_{a3} = \frac{1}{V_3} \cdot \dot{V} = \frac{1}{14.25 \frac{\text{ft}^3}{\text{lb}_a}} \cdot 8.33 \frac{\text{ft}^3}{\text{s}}$$

$$V_3 = 14.25 \frac{\text{ft}^3}{\text{lb}_a} \quad \dot{V} = 8.33 \frac{\text{ft}^3}{\text{s}}$$

$$\dot{m}_{a3} = 0.585 \frac{\text{lb}_a}{\text{s}}$$

$$\dot{m}_{a4} = 3.83 \frac{\text{lb}_a}{\text{s}}$$

$$\dot{m}_{a4} + \dot{m}_{a3} = 4.42 \frac{\text{lb}_a}{\text{s}}$$

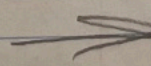
Mixing box $\rightarrow \frac{\dot{m}_{a4}}{\dot{m}_{a3}} = \frac{w_0 - w_1}{w_1 - w_4} \rightarrow w_1 = \frac{\dot{m}_{a4} \cdot w_4 + \dot{m}_{a3} \cdot w_0}{\dot{m}_{a4} + \dot{m}_{a3}}$

$$w_4 = 0.009 \frac{\text{lb}_w}{\text{lb}_a} \quad w_0 = 0.018 \frac{\text{lb}_w}{\text{lb}_a}$$

$$w_1 = \frac{(3.83 \frac{\text{lb}_a}{\text{s}} \cdot 0.009 \frac{\text{lb}_w}{\text{lb}_a}) + (0.585 \frac{\text{lb}_a}{\text{s}} \cdot 0.018 \frac{\text{lb}_w}{\text{lb}_a})}{4.42 \frac{\text{lb}_a}{\text{s}}}$$

$$w_1 = 0.01026$$

Sources:
thermodynamics
an engineering
approach
Ayala
Notes/slides



sources
 - Ayala
 - thermodynamics
 an engineering
 approach
 - notes/slides

4

calculations cont.

Dehumidifier

1st law
 (1-1)

$$\dot{Q}_{out} = \dot{m}_a (h_1 - h_2) + \dot{m}_a (w_1 - w_2) h_w$$

$$= \frac{4.44 \text{ lbm}}{\text{s}} (29.5 - 23) \frac{\text{Btu}}{\text{lbm}} + \frac{4.44 \text{ lbm}}{\text{s}} (0.01026 - 0.0091) \frac{\text{lbm}}{\text{lbm}} \times 23.07 \frac{\text{Btu}}{\text{lbm}}$$

$h_w = 23.07 \frac{\text{Btu}}{\text{lbm}}$
 $h_1 = 29.5 \frac{\text{Btu}}{\text{lbm}}$
 $h_2 = 23 \frac{\text{Btu}}{\text{lbm}}$

$\dot{Q}_{out} = 28.978 \frac{\text{Btu}}{\text{s}}$

Solution 5: Problem 1

- | | | | |
|---|--|--|--|
| <p>①</p> <p>$T_{db1} = 76.5^\circ\text{F}$</p> <p>$w_1 = 0.01026 \frac{\text{lbm}}{\text{lbm}}$</p> | <p>②</p> <p>$T_{db1} = 55^\circ\text{F}$</p> <p>$w_1 = 0.0091 \frac{\text{lbm}}{\text{lbm}}$</p> | <p>③</p> <p>$T_{db3} = 55^\circ\text{F}$</p> <p>$w_3 = 0.0091 \frac{\text{lbm}}{\text{lbm}}$</p> | <p>④</p> <p>$T_{db4} = 75^\circ\text{F}$</p> <p>$w_4 = 0.0091 \frac{\text{lbm}}{\text{lbm}}$</p> |
| <p>⑤</p> <p>$T_{db5} = 75^\circ\text{F}$</p> <p>$w_5 = 0.0091 \frac{\text{lbm}}{\text{lbm}}$</p> | <p>⑥</p> <p>$T_{db} = 90^\circ\text{F}$</p> <p>$w_0 = 0.018 \frac{\text{lbm}}{\text{lbm}}$</p> | | |

Technical analysis:

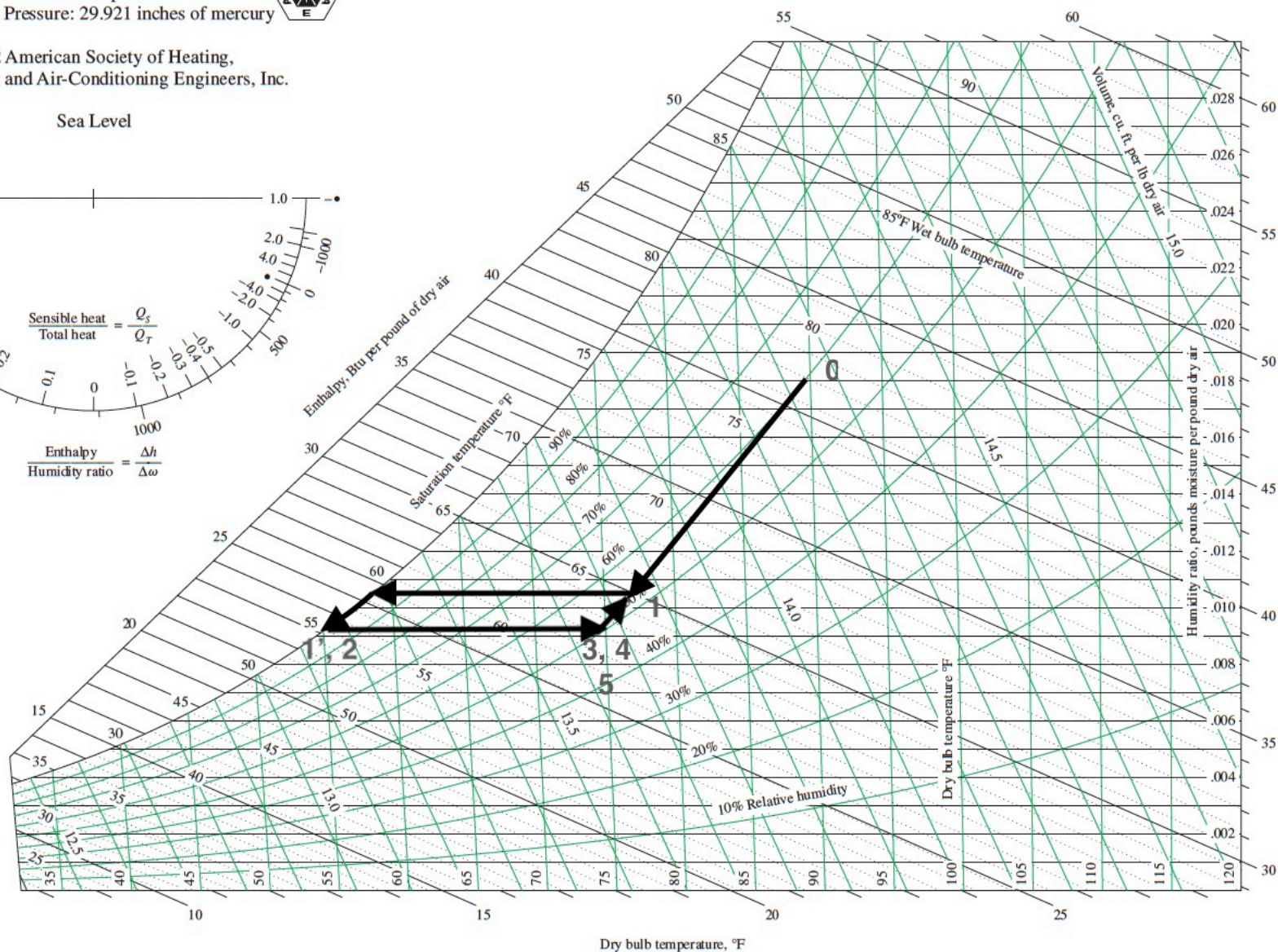
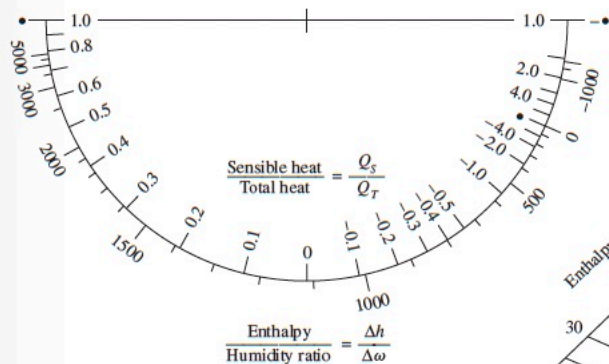
Based on the system as it stands it will function fine, but, with changes to the outside temp., the heat gained inside the room or a change to the cooling and dehumidifying unit, the system may break or not function. For example, if the outside temp and humidity change the air at the mixing chamber may be too cool to enter the rest of the system making the temp. too cold. if the Qin changes, the room may be too warm. This system relies on a balance and does not seem very adaptive, in case of these issues, a possible fix could be adding in a temp. sensitive humidifier in case it gets too cold, or if it is too hot, increasing the efficacy of the cooling and dehumidifying unit to expel more heat.

ASHRAE Psychrometric Chart No. 1
Normal Temperature
Barometric Pressure: 29.921 inches of mercury



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Sea Level



Prepared by Center for Applied Thermodynamic Studies, University of Idaho.

FIGURE A-31E

Psychrometric chart at 1 atm total pressure.

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