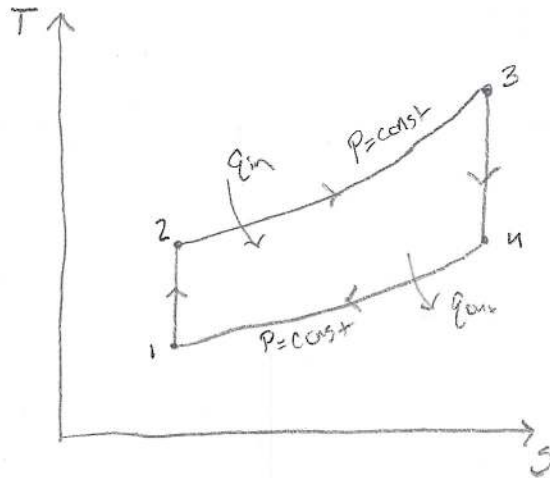


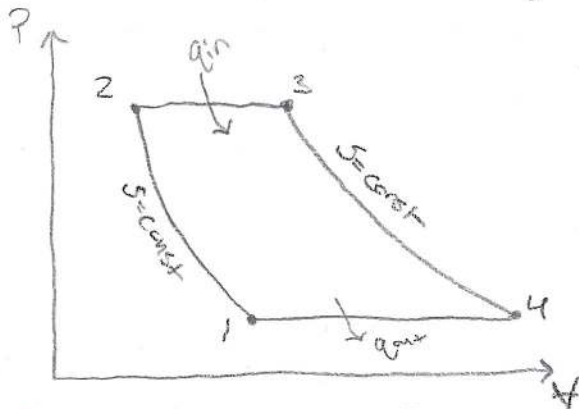
## Problem 9-88

An aircraft engine operates on a simple ideal Brayton cycle with a pressure ratio of 10. Heat is added to the cycle at a rate of 500 kW and the air passes through the engine at a rate of 1 kg/s, and the air at the beginning of the compression is at 70 kPa and 0°C. Determine the power produced by this engine and its thermal efficiency. (Use constant specific heats at room temperature.)



Given:

$$\begin{aligned} r_p &= 10 \\ \dot{Q}_{in} &= 500 \text{ kW} \\ \dot{m}_{air} &= 1 \text{ kg/s} \\ P_1 &= 70 \text{ kPa} \\ T_1 &= 0^\circ\text{C} = 273 \text{ K} \end{aligned}$$



Process 1-2) Isentropic

$$r_p = \left( \frac{P_2}{P_1} \right) = 10 \quad 10 = \left( \frac{P_2}{70 \text{ kPa}} \right) \quad P_2 = 700 \text{ kPa}$$

$$\frac{T_2}{T_1} = \left( \frac{P_2}{P_1} \right)^{\frac{k-1}{k}} \quad T_2 = 273 (10)^{\frac{1.4-1}{1.4}} = 527.1 \text{ K}$$

Process 2 → 3)  $P = \text{const}$

$$P_2 = P_3 \quad P_3 = 700 \text{ kPa}$$

$$\dot{Q}_{in} = 500 \text{ kW}$$

$$\dot{Q}_{in} = C_p \cdot (T_3 - T_2) \cdot \dot{m}$$

$$500 \text{ kW} \left( \frac{1 \text{ kJ/s}}{1 \text{ kW}} \right) = 1,005 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} (T_3 - 527.1 \text{ K}) \cdot 1 \frac{\text{kg}}{\text{s}}$$

$$\frac{500 \text{ kJ/s}}{1 \text{ kg/s}} = 1,005 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} (T_3 - 527.1 \text{ K})$$

$$\frac{500 \frac{\text{kJ}}{\text{s}}}{1,005 \frac{\text{kJ}}{\text{kg} \cdot \text{K}}} = (T_3 - 527.1 \text{ K})$$

$$497.61 \text{ K} + 527.1 \text{ K} = T_3 = 1024.61 \text{ K}$$

Process 3 → 4) Isentropic expansion

$$\frac{T_3}{T_4} = \left( \frac{P_3}{P_4} \right)^{\frac{\gamma-1}{\gamma}} \quad T_4 = \frac{T_3}{(r_p)^{\frac{\gamma-1}{\gamma}}}$$

$$T_4 = \frac{1024.61}{(10)^{\frac{1.4-1}{1.4}}} = 358.61 \text{ K}$$

$$P_4 = P_1 \quad P_4 = 70 \text{ kPa}$$

Calculating Questions)

$$\dot{Q}_{in} = 500 \text{ kJ}$$

$$\dot{Q}_{out} = C_p (T_4 - T_1) \cdot \dot{m}$$

$$= 1,005 \frac{\text{kJ}}{\text{kg} \cdot \text{K}} (358.61 \text{ K} - 272 \text{ K}) \cdot 1 \frac{\text{kg}}{\text{s}}$$

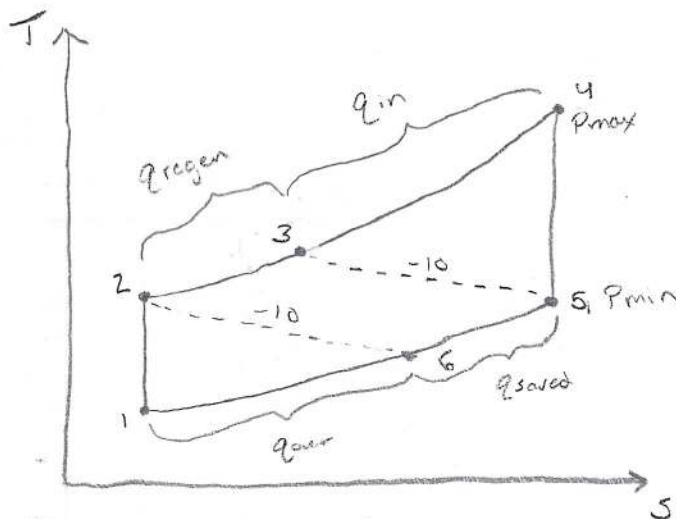
$$\dot{Q}_{out} = 86.04 \text{ kJ}$$

$$W_{net} = \dot{Q}_{in} - \dot{Q}_{out} = 500 \text{ kJ} - 86.04 \text{ kJ} = \boxed{413.96 \text{ kW}}$$

$$\eta = \frac{W_{net}}{\dot{Q}_{in}} \cdot 100\% = \frac{413.96}{500} = \boxed{82.8\%}$$

## Problem 9-99

A gas turbine for an automobile is designed with a regenerator. Air enters the compressor of this engine at 100 kPa and 30°C. The compressor pressure ratio is 10; the maximum cycle temperature is 800°C; and the cold air stream leaves the regenerator 10°C cooler than the hot air stream at the inlet of the regenerator. Assuming both the compressor and the turbine to be isentropic, determine the rates of heat addition and rejection for this cycle when it produces 115 kW. (Use constant specific heats at room temp)



Given:

$$T_1 = 30^\circ\text{C} = 303\text{ K}$$

$$P_1 = 100\text{ kPa}$$

$$r_p = 10$$

$$T_4 = 800^\circ\text{C} = 1073\text{ K}$$

$$T_5 - T_3 = 10^\circ\text{C}$$

$$T_6 - T_2 = 10^\circ\text{C}$$

Process 1 → 2) Isentropic

$$T_1 = 303\text{ K} \quad P_1 = 100\text{ kPa}$$

$$\frac{P_2}{P_1} = r_p = 10$$

$$\frac{T_2}{T_1} = (r_p)^{\frac{k-1}{k}}$$

$$T_2 = 303\text{ K} (10)^{\frac{1.4-1}{1.4}}$$

$$T_2 = 585\text{ K}$$

$$\frac{P_2}{P_1} = 10$$

$$P_2 = 100(10) = 1000\text{ kPa}$$

Process 4 → 5) Isentropic

$$T_4 = 1073\text{ K}$$

$$\frac{T_4}{T_5} = \left(\frac{P_4}{P_5}\right)^{\frac{k-1}{k}}$$

$$\frac{T_4}{(r_p)^{\frac{k-1}{k}}} = T_5 = \frac{1073\text{ K}}{(10)^{\frac{1.4-1}{1.4}}} = 555.76\text{ K}$$

$$P_2 = P_4 = 1000\text{ kPa}$$

$$P_1 = P_5 = 100\text{ kPa}$$

— Finding  $T_3$  &  $T_6$

$$T_3 = T_5 - 10K = 555.76 - 10 = 545.76 K$$

$$T_6 = 10K + T_2 = 585K + 10K = 595K$$

— Finding  $Q_{in}$  &  $Q_{out}$

$$W_{net} = W_T - W_C$$

$$W_C = c_p (T_2 - T_1) \dot{m}$$

$$W_T = c_p (T_4 - T_5) \dot{m}$$

$$W_{net} = c_p (T_4 - T_5) \dot{m} - c_p (T_2 - T_1) \dot{m}$$

$$115 KJ = c_p (1073K - 555.76) \dot{m} - c_p (585K - 303K)$$

$$115 KJ = c_p \cdot \dot{m} (517.24 K - 282 K)$$

$$115 KJ = 1.005 \frac{KJ}{kg \cdot K} \cdot \dot{m} (235.24 K)$$

$$\dot{m} = 0.4864 kg$$

$$Q_{in} = \dot{m} \cdot c_p (T_4 - T_3)$$

$$= 0.4864 kg \cdot 1.005 \frac{KJ}{kg \cdot K} (1073K - 545.76K)$$

$$Q_{in} = 257.7 KJ = \boxed{257.7 kW}$$

$$Q_{out} = \dot{m} \cdot c_p (T_6 - T_1)$$

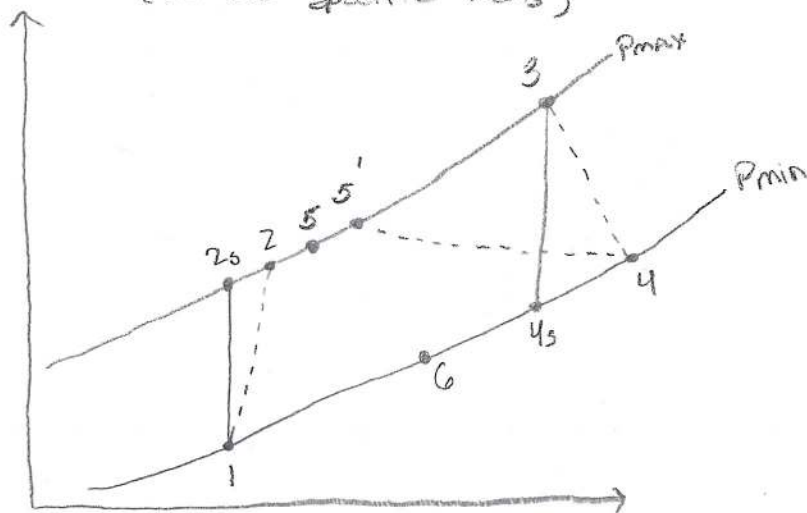
$$= 0.4864 kg \cdot 1.005 \frac{KJ}{kg \cdot K} (595K - 303K)$$

$$Q_{out} = 142.74 KJ = \boxed{142.74 kW}$$

## Problem 9-107

A Brayton cycle with regeneration using air as the working fluid has a pressure ratio of 7. The minimum and maximum temperatures in the cycle are 310 and 1150 K. Assuming an isentropic efficiency of 75 percent for the compressor and 82 percent for the turbine and an effectiveness of 65 percent for the regenerator, determine (a) the temperature at the turbine exit, (b) the network output, and (c) the thermal efficiency.

(Variable specific heats)



Given:-

$$r_p = 7$$

$$T_1 = 310 \text{ K}$$

$$T_3 = 1150 \text{ K}$$

$$\eta_c = 0.75$$

$$\eta_T = 0.82$$

$$\varepsilon = 0.65$$

Process 1  $\rightarrow$  2) Isentropic

$$T_1 = 310 \text{ K}$$

$$P_{r1} = 1.5546$$

$$h_1 = 310.24 \text{ kJ/kg}$$

$$\frac{P_{r2s}}{P_{r1}} = \frac{P_2}{P_1} = r_p$$

$$P_{r2s} = P_{r1} (r_p) = 1.5546 (7) = 10.8822$$

$P_r$	
10.37	533.98
10.8822	$h_{2s}$
11.10	544.35

$$\frac{10.8822 - 10.37}{11.10 - 10.37} = \frac{h_{2s} - 533.98}{544.35 - 533.98}$$

$$h_{2s} = 541.26 \text{ kJ/kg}$$

$$\eta_c = \frac{h_{2s} - h_1}{h_2 - h_1}$$

$$h_2 = \frac{h_{2s} - h_1}{\eta_c} + h_1 = \frac{541.26 - 310.24}{0.75} + 310.24$$

$$h_2 = 618.27 \text{ kJ/kg}$$

Process 3 → 4) Isentropic

$$T_3 = 1150 \text{ K}$$

T	Pr
1140	193.1
1150	Pr <sub>3</sub>
1160	207.2

$$\frac{1150 - 1140}{1160 - 1140} = \frac{Pr_3 - 193.1}{207.2 - 193.1}$$

$$Pr_3 = 200.15$$

T	h
1140	1207.57
1150	h <sub>3</sub>
1160	1230.92

$$\frac{1150 - 1140}{1160 - 1140} = \frac{h_3 - 1207.57}{1230.92 - 1207.57}$$

$$h_3 = 1219.25 \text{ kJ/kg}$$

$$\frac{P_3}{P_{4s}} = \frac{Pr_3}{Pr_{4s}} = r_p \quad Pr_{4s} = \frac{Pr_3}{r_p} = \frac{200.15}{7} = 28.59$$

Pr	h
27.29	702.52
28.59	h <sub>4s</sub>
28.80	713.27

$$\frac{28.59 - 27.29}{28.80 - 27.29} = \frac{h_4 - 702.52}{713.27 - 702.52}$$

$$h_{4s} = 711.77 \text{ kJ/kg}$$

$$\eta_T = \frac{h_3 - h_4}{h_3 - h_{4s}} \quad h_4 = \eta_T (h_3 - h_{4s}) - h_3 \cdot -1$$

$$h_4 = -(0.82 (1219.25 - 711.77) - 1219.25) = 803.12 \text{ kJ/kg}$$

Process 4 → 5)  $\epsilon = 0.65$

$$\epsilon = \frac{q_{regen}}{q_{regen\ max}} = \frac{h_5 - h_2}{h_4 - h_2} \quad \epsilon (h_4 - h_2) + h_2 = h_5$$

$$h_5 = 0.65 (803.12 - 618.27) + 618.27 = 738.42 \text{ kJ/kg}$$

- calculating Questions

a) T at turbine exit

T	h
780	800.03
$T_4$	803.12
800	821.95

$$\frac{T_4 - 780}{800 - 780} = \frac{803.12 - 800.03}{821.95 - 800.03}$$

$$T_4 = \boxed{782.81 \text{ K}}$$

b)  $W_{net}$

$$W_{net} = W_T - W_C = (h_3 - h_4) - (h_2 - h_1)$$

$$W_{net} = (1219.25 - 803.12) - (618.27 - 310.24) = \boxed{108.1 \text{ kJ/kg}}$$

c)  $\eta_{Th}$

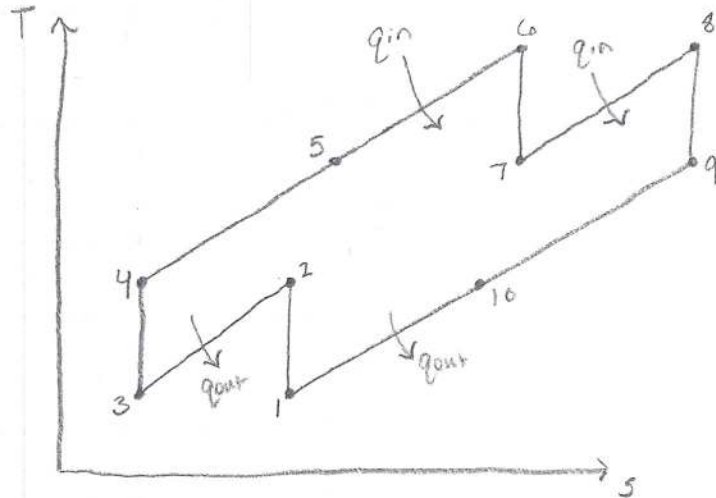
$$\eta_{Th} = \frac{W_{net}}{q_{in}}$$

$$q_{in} = h_3 - h_5$$

$$\eta_{Th} = \frac{108.1}{(1219.25 - 738.42)} = \boxed{22.5 \%}$$

## Problem 9-119

Consider a regenerative gas-turbine power plant with two stages of compression and two stages of expansion. The overall pressure ratio of the cycle is 9. The air enters each stage of the compressor at 300 K and each stage of the turbine at 1200 K. Accounting for the variation of specific heats with temperature, determine the minimum mass flow rate of the air needed to develop a net power of 110 MW.



Given:

$$r_p = 9$$

$$T_1 = T_3 = 300 \text{ K}$$

$$T_6 = T_8 = 1200 \text{ K}$$

$$\frac{P_2}{P_1} = \frac{P_4}{P_3} = \sqrt{9} = 3 \quad \frac{P_6}{P_7} = \frac{P_8}{P_9} = \sqrt{9} = 3$$

— Values for  $T_1$  &  $T_3$ 

$$T_1 = T_3 = 300 \text{ K}$$

$$h_1 = h_3 = 300.19 \text{ kJ/kg}$$

$$Pr_1 = 1.3860$$

— Values for  $T_2$  &  $T_4$ 

$$\frac{Pr_2}{Pr_1} = \frac{P_2}{P_1} \quad Pr_2 = Pr_1 \left( \frac{P_2}{P_1} \right) \quad \frac{P_2}{P_1} = 3$$

$$Pr_2 = 1.3860 (3) = 4.158$$

$Pr$	$T$
4.153	410
4.158	$T_2$
4.522	420

$$\frac{4.158 - 4.153}{4.522 - 4.153} = \frac{T_2 - 410}{420 - 410}$$

$$T_2 = T_4 = 410.13 \text{ K}$$

$P_r$	$h$
4.153	411.12
4.158	$h_2$
4.522	421.26

$$\frac{4.158 - 4.153}{4.522 - 4.153} = \frac{h_2 - 411.12}{421.26 - 411.12}$$

$$h_2 = h_u = 411.26 \text{ kJ/kg}$$

— Values for  $T_6$  &  $T_8$

$$T_6 = T_8 = 1200 \text{ K}$$

$$h_6 = h_8 = 1277.79 \text{ kJ/kg}$$

$$P_{r6} = 238.0$$

— Values for  $T_7$  &  $T_9$

$$\frac{P_{r6}}{P_{r7}} = \frac{P_6}{P_7} = r_p$$

$$P_{r7} = \frac{P_{r6}}{r_p} = \frac{238}{3} = 79.33$$

$P_r$	$T$
75.29	900
79.33	$T_7$
82.05	920

$$\frac{79.33 - 75.29}{82.05 - 75.29} = \frac{T_7 - 900}{920 - 900}$$

$$T_7 = T_9 = 911.95 \text{ K}$$

$P_r$	$h$
75.29	932.93
79.33	$h_7$
82.05	955.38

$$\frac{79.33 - 75.29}{82.05 - 75.29} = \frac{h_7 - 932.93}{955.38 - 932.93}$$

$$h_7 = h_9 = 946.35 \text{ kJ/kg}$$

— Calculating questions

$$W_{net} = W_T - W_C = (2(h_6 - h_7)) - (2(h_4 - h_3))$$

$$W_{net} = (2(1277.79 - 946.35)) - (2(411.26 - 300.19))$$

$$W_{net} = 440.74 \text{ kJ/kg}$$

$$\dot{W} = 110 \text{ MW} = 110,000 \text{ kJ/s}$$

$$\dot{W} = \dot{m} w_{\text{net}}$$

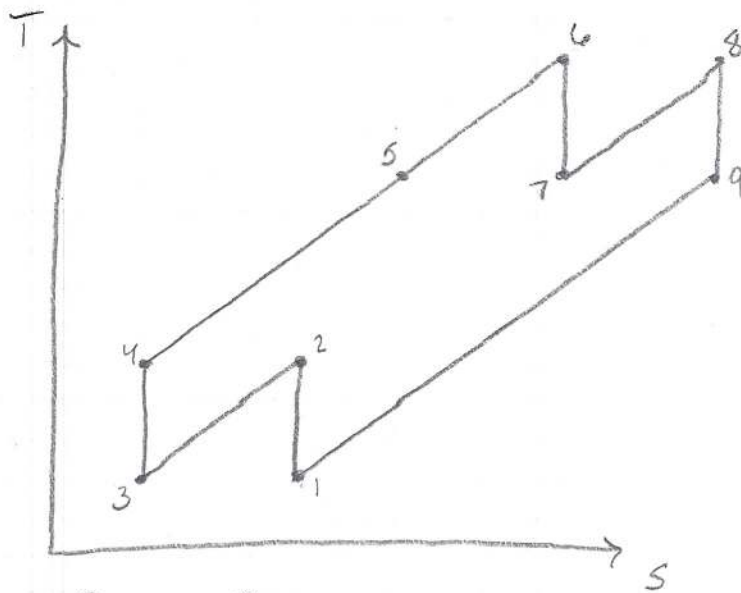
$$110,000 \text{ kJ/s} = \dot{m} (440.74 \text{ kJ/kg})$$

$$\dot{m} = 249.58 \text{ kg/s}$$

## Problem 9-121

Consider an Ideal gas-turbine cycle with two stages of compression and two stages of expansion. The pressure ratio across each stage of the compressor and turbine is 3. The air enters each stage of the compressor at 300K and each stage of the turbine at 1200K. Determine the back work ratio and thermal efficiency of the cycle assuming (a) no regenerator is used and (b) a regenerator with 75% effectiveness is used.

(Variable specific heats)



Given:

$$r_p = 3$$

$$T_1 = T_3 = 300 \text{ K}$$

$$T_6 = T_8 = 1200 \text{ K}$$

$$\frac{P_2}{P_1} = \frac{P_4}{P_3} = 3$$

$$\frac{P_6}{P_7} = \frac{P_8}{P_9} = 3$$

Values for  $T_1$  &  $T_3$

$$T_1 = T_3 = 300 \text{ K}$$

$$h_1 = h_3 = 300.19 \text{ kJ/kg}$$

$$P_{r1} = 1.3860$$

— Values for  $T_2$  &  $T_4$

$$\frac{P_{r2}}{P_{r1}} = \frac{P_2}{P_1} = r_p \quad P_{r2} = P_{r1}(r_p)$$

$$P_{r2} = (1.3860)(3) = 4.158$$

$P_r$	$h$	
4.153	411.12	$\frac{4.158 - 4.153}{4.522 - 4.153} = \frac{h_2 - 411.12}{421.26 - 411.12}$
4.158	$h_2$	
4.522	421.26	

$h_2 = h_4 = 411.26 \text{ kJ/kg}$

— Values for  $T_6$  &  $T_8$

$$T_6 = T_8 = 1200 \text{ K}$$

$$h_6 = h_8 = 1277.79 \text{ kJ/kg}$$

$$P_{r6} = 238.0$$

— Values for  $T_7$  &  $T_9$

$$\frac{P_{r6}}{P_{r7}} = \frac{P_6}{P_7} = r_p \quad P_{r7} = \frac{P_{r6}}{r_p} = \frac{238}{3} = 79.33$$

$P_r$	$h$	
75.29	932.93	$\frac{79.33 - 75.29}{82.05 - 75.29} = \frac{h_7 - 932.93}{955.38 - 932.93}$
79.33	$h_7$	
82.05	955.38	

$h_7 = h_9 = 946.35 \text{ kJ/kg}$

— Calculating Back work (No Regen)

$$r_{BW} = \frac{W_C}{W_T} = \frac{2(h_4 - h_3)}{2(h_6 - h_7)}$$

$$\frac{2(411.26 - 300.19)}{2(1277.79 - 946.35)} = \boxed{0.335}$$

— Calculating Thermal efficiency (No Regen)

$$\eta_{Th} = \frac{W_{net}}{q_{in}}$$

$$W_{net} = W_T - W_C = (2(h_6 - h_7)) - (2(h_4 - h_3))$$

$$W_{net} = (2(1277.79 - 946.35)) - (2(411.26 - 300.19))$$

$$W_{net} = 440.74 \text{ kJ/kg}$$

$$q_{in} = (h_6 - h_4) + (h_8 - h_7) = 1197.97 \text{ kJ/kg}$$

$$q_{in} = (1277.79 - 411.26) + (1277.79 - 946.35) = 1197.97 \text{ kJ/kg}$$

$$\eta_{Th} = \frac{440.74}{1197.97} = \boxed{36.8\%}$$

Calculating Backwork (Regen Added)

NO change to turbine or compressor

$$\boxed{r_{bw} = 0.335}$$

Calculating Thermal efficiency (Regen Added)

$$q_{regen} = 0.75(h_9 - h_4) = 0.75(946.35 - 411.26)$$

$$q_{regen} = 401.32 \text{ kJ/kg}$$

$$q_{in} = q_{in,old} - q_{regen} = 1197.97 - 401.32$$

$$q_{in} = 796.65 \text{ kJ/kg}$$

$$\eta_{Th} = \frac{440.74}{796.65} = \boxed{55.3\%}$$