

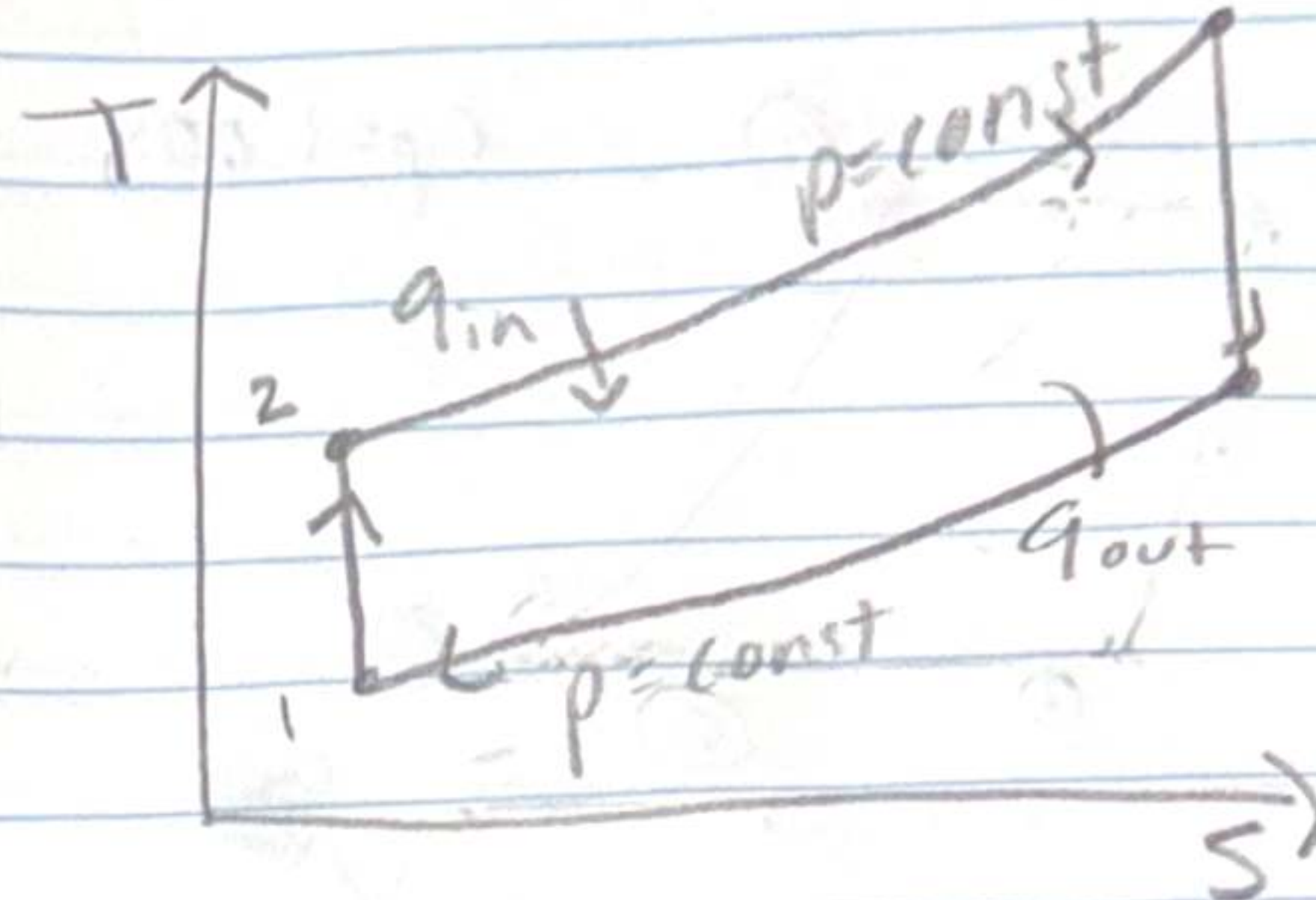
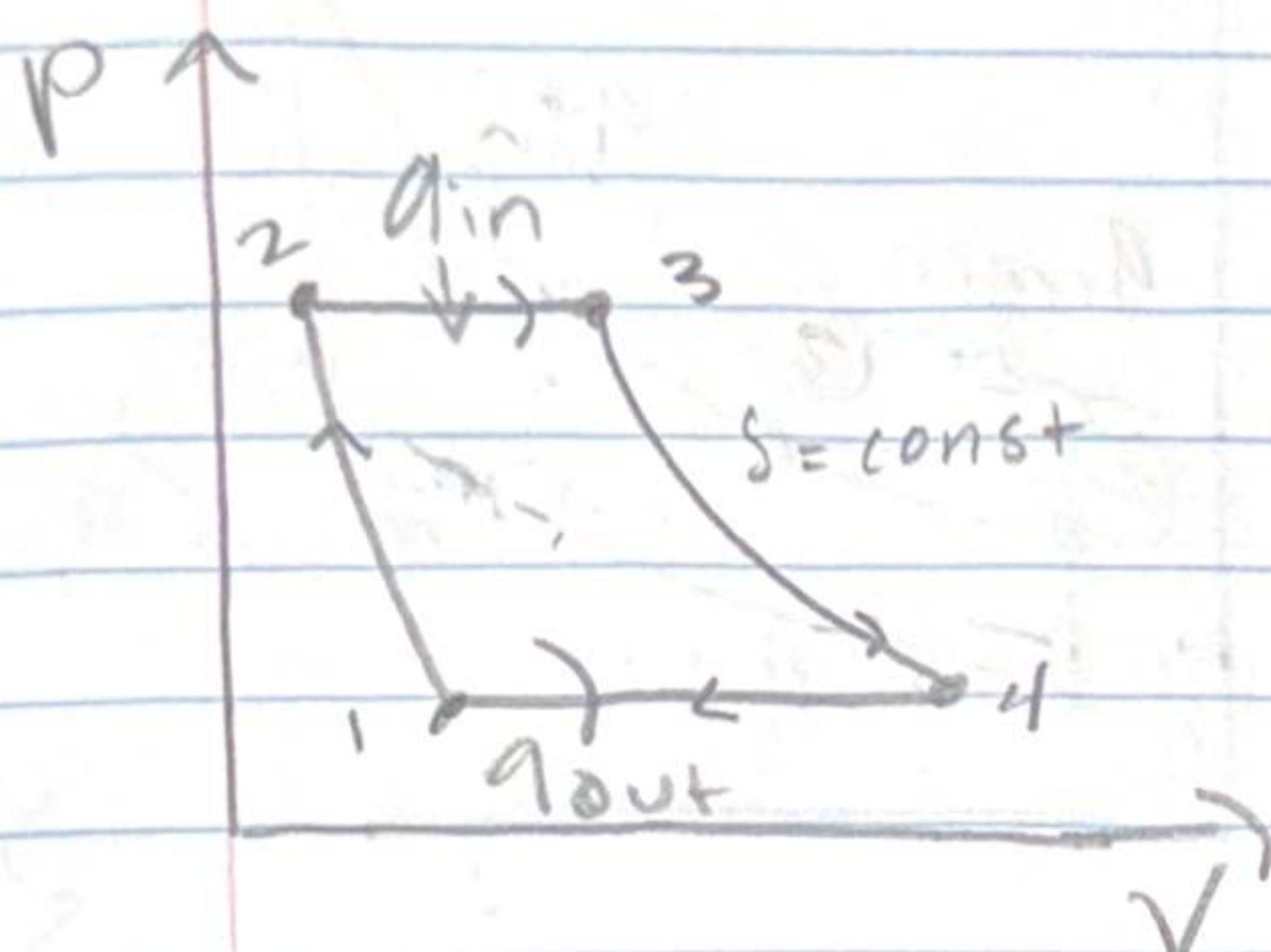
HW 1.4

9-886

Ideal Brayton Cycle

Mizell
smalls

$$r_p = 10 \quad k = 1.4 \quad \eta = .287 \quad C_p = 1.005 \quad C_v = .718$$



$$q_{in} = 500 \text{ kW}$$

$$\dot{m} = 1 \text{ kg/s}$$

① Isentropic ② p const ③ Isentropic ④

Constant p

$$P_1 = 70 \text{ kPa} \quad P_2 = 700 \text{ kPa} \quad P_3 = 700 \text{ kPa} \quad P_4 = 70 \text{ kPa}$$

$$T_1 = 273 \text{ K} \quad T_2 = 527 \text{ K} \quad T_3 = 1024 \text{ K} \quad T_4 = 530 \text{ K}$$

$$V_1 = 1.119 \quad V_2 = 0.216 \quad V_3 = 0.419 \quad V_4 = 2.173$$

$$P_2 = r_p \times P_1$$

$$T_2 = T_1 \left(\frac{P_2}{P_1} \right)^{(k-1)/k}$$

$$T_4 = T_3 \left(\frac{P_4}{P_3} \right)^{(k-1)/k}$$

$$\frac{k-1}{k} = .2857$$

$$T_3 = T_2 + \frac{q_{in}}{C_p}$$

$$W_{net} = C_p [(T_3 - T_2) - (T_4 - T_1)]$$

$$= 1.005 [(497) - (267)]$$

$$W_{net} = 241.2$$

$$\dot{W}_{net} = \dot{m} \cdot W_{net}$$

$$= 1 \cdot 241.2$$

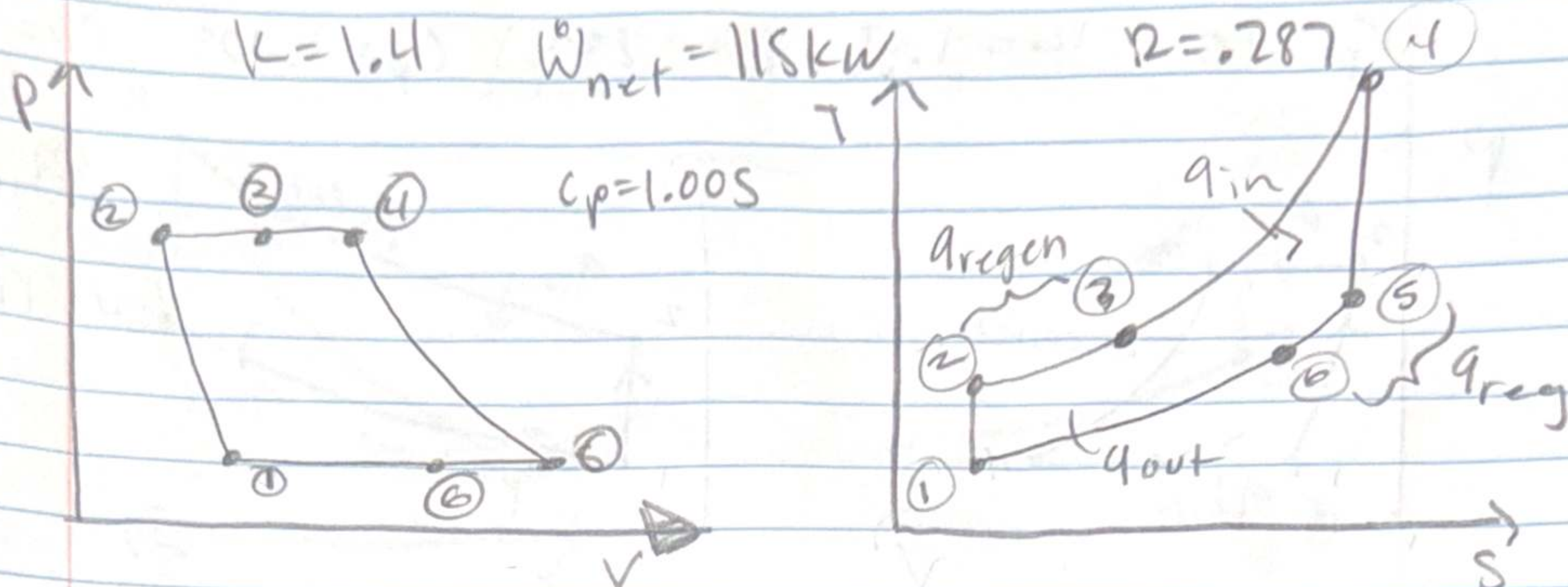
$$W_{net} = 241.2$$

$$\eta_{th} = \frac{241.2}{500 \text{ kW}} \rightarrow 0.4824$$

$$pV = kT$$

$$V = \frac{RT}{p}$$

9-99 | $r_p = 10$ $T_{max} = 800^\circ\text{C} \rightarrow 1073\text{K}$



① T_{sent} ② P_{const} ③ P_{const} ④ T_{sent} ⑤ P_{const} ⑥ P_{const}

$P_1 = 100\text{ kPa}$	$P_2 = 1000$	$P_3 = 1000$	$P_4 = 1000$	$P_5 = 100\text{ kPa}$
$T_1 = 303\text{ K}$	$T_2 = 884.9$	$T_3 = 845.7$	$T_4 = 1073\text{ K}$	$T_5 = 885.7\text{ K}$
$V_1 = .86$	$V_2 = .167$	$V_3 = .186$	$V_4 = .307$	$V_5 = 1.59$

$PV = RT$

$\frac{k-1}{k}$
 $= .2857$

$V = \frac{RT}{P}$

$\frac{T_2}{T_1} = \left(\frac{P_2}{P_1}\right)^{\frac{k-1}{k}}$

$T_3 = T_5 - 10$

$P_2 = P_3 = P_4$

$\frac{T_5}{T_4} = \left(\frac{P_5}{P_4}\right)^{\frac{k-1}{k}}$

$P_6 = 100\text{ kPa}$
 $T_6 = 894.9\text{ K}$
 $V_6 = 1.707$

$P_6 = P_1$

$q_{in} = c_p(T_4 - T_3)$
 $= 1.005(1073 - 845.7)$
 $= 829.93$

$\dot{W}_{net} = \dot{m}(q_{in} - q_{out})$

$q_{out} = c_p(T_6 - T_1)$
 $= 1.005(894.9 - 303)$
 $= 293.35$

$\dot{m} = 115$
 $829.9 - 293.35$

$\dot{m} = .486$

$\dot{Q}_{out} = \dot{m}q_{out}$
 $= (.486)(293.35)$
 $= 142.6681\text{ kW}$

$\dot{Q}_{in} = \dot{m}q_{in}$
 $= (.486)(829.93)$
 $= 257.841\text{ kW}$

$$9-107 \quad r_p = 7$$

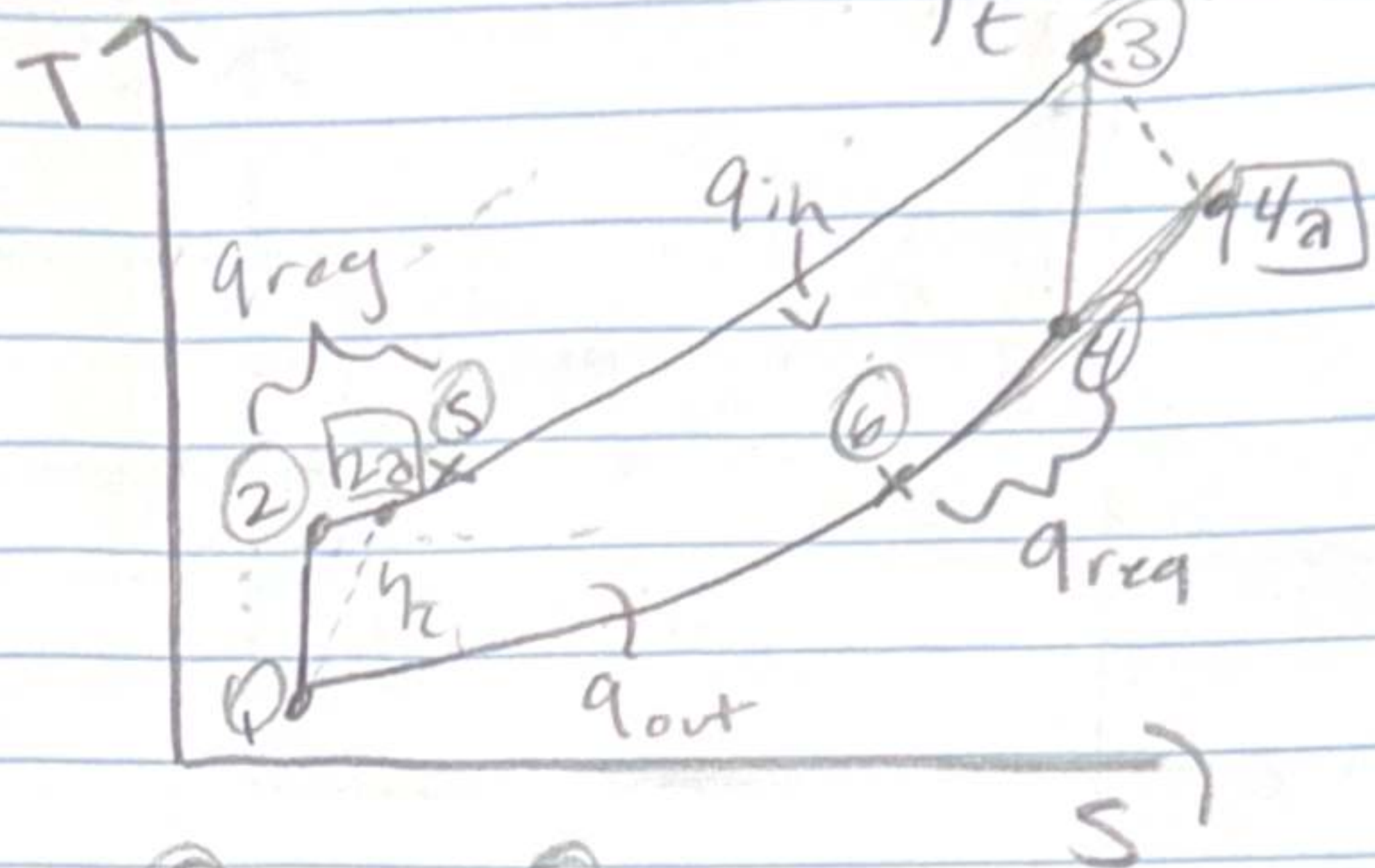
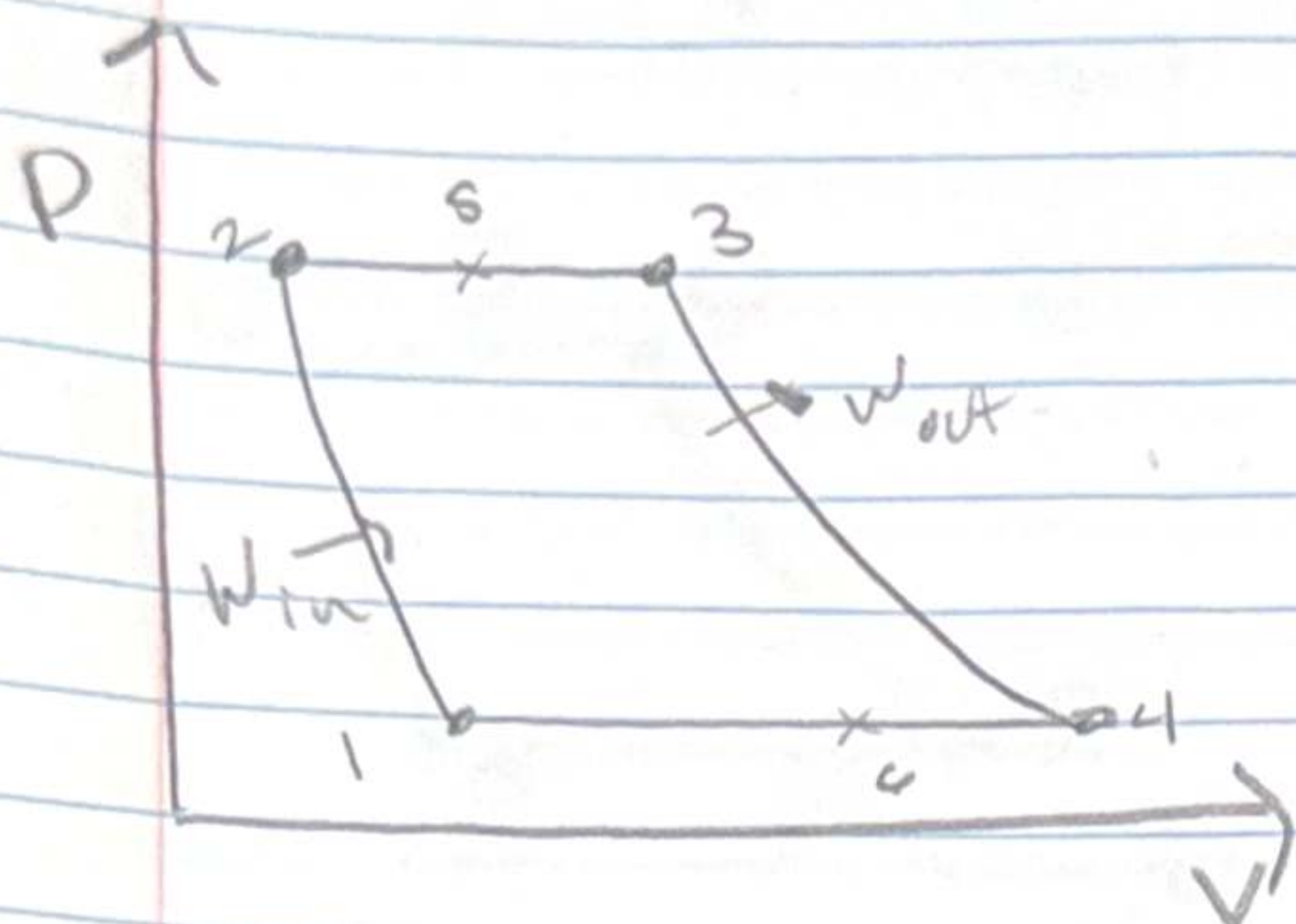
$$T_{min} = 310K$$

$$T_{max} = 1150K$$

$$\epsilon = 0.65$$

$$\eta_c = 0.75$$

$$\eta_t = 0.82$$



$$h_{4a} = 803.14$$

Using Table A-17	$T_1 = 310K$	$P_2 = 7P_1$	$h_5 = 738.4$	$T_3 = 1150K$	$h_4 = 711.7$	$P_6 = P_1$
	$P_r = 1.854$	$T_2 = 837$	$P_5 = P_2$	$P_3 = P_2$	$P_4 = P_1$	$T_6 = 671.71$
	$h_1 = 310.24$	$h_{2a} = 618.2$	$T_5 = 723.3$	$P_{r3} = 200.15$	$P_{r4} = 28.59$	$h_6 = 682.97$
	$T_2 = \left(\frac{P_2}{P_1}\right)^{\frac{k-1}{k}}$	$T_{2a} = 610.7$	$\epsilon = T_5 - T_{2a}$	$T_4 = \left(\frac{P_4}{P_3}\right)^{\frac{k-1}{k}}$		

$$\frac{P_2}{P_1} = \frac{P_{r2}}{P_{r1}} \Rightarrow P_{r2} = 10.82 \Rightarrow T_2$$

$$T_5 = 702.16K \quad P_{r3} \Rightarrow A-17$$

$$A) T_{Turb, out} = T_{4a} = 747.82K \quad A-17$$

$$P_{r4} = 28.513$$

$$T_{4a} = 698.63K \quad A-17$$

$$B) W_{net} = C_p(T_3 - T_{4a}) - C_p(T_{2a} - T_1)$$

$$C) \eta_{th} = \frac{W_{net}}{q_{in}}, \quad q_{in} = C_p(T_3 - T_5)$$

$$\eta_c = \frac{h_2 - h_1}{h_{2a} - h_1} \Rightarrow h_{2a}$$

$$\epsilon = \frac{h_5 - h_{2a}}{h_{4a} - h_{2a}}$$

$$h_5 = 738.4$$

$$A-17$$

$$T_5 = 723.3$$

$$h_3 \rightarrow A-17$$

$$A-17$$

$$T_{4a} = 782.84$$

$$T_{2a} \Rightarrow A-17 \quad P_5 = P_2$$

9-107 (cont'd)

$$W_{net} = W_{out} - W_{in}$$

$$W_{out} = h_3 - h_{4a} = 416.10$$

$$W_{in} = h_2 - h_1 = 308.03$$

$$\boxed{W_{net} = 108.075}$$

$$c) \eta_{th} = \frac{W_{net}}{q_{in}}$$

$$q_{in} = h_3 - h_g = 480.81$$

$$\eta_{th} = \frac{108.075}{480.81}$$

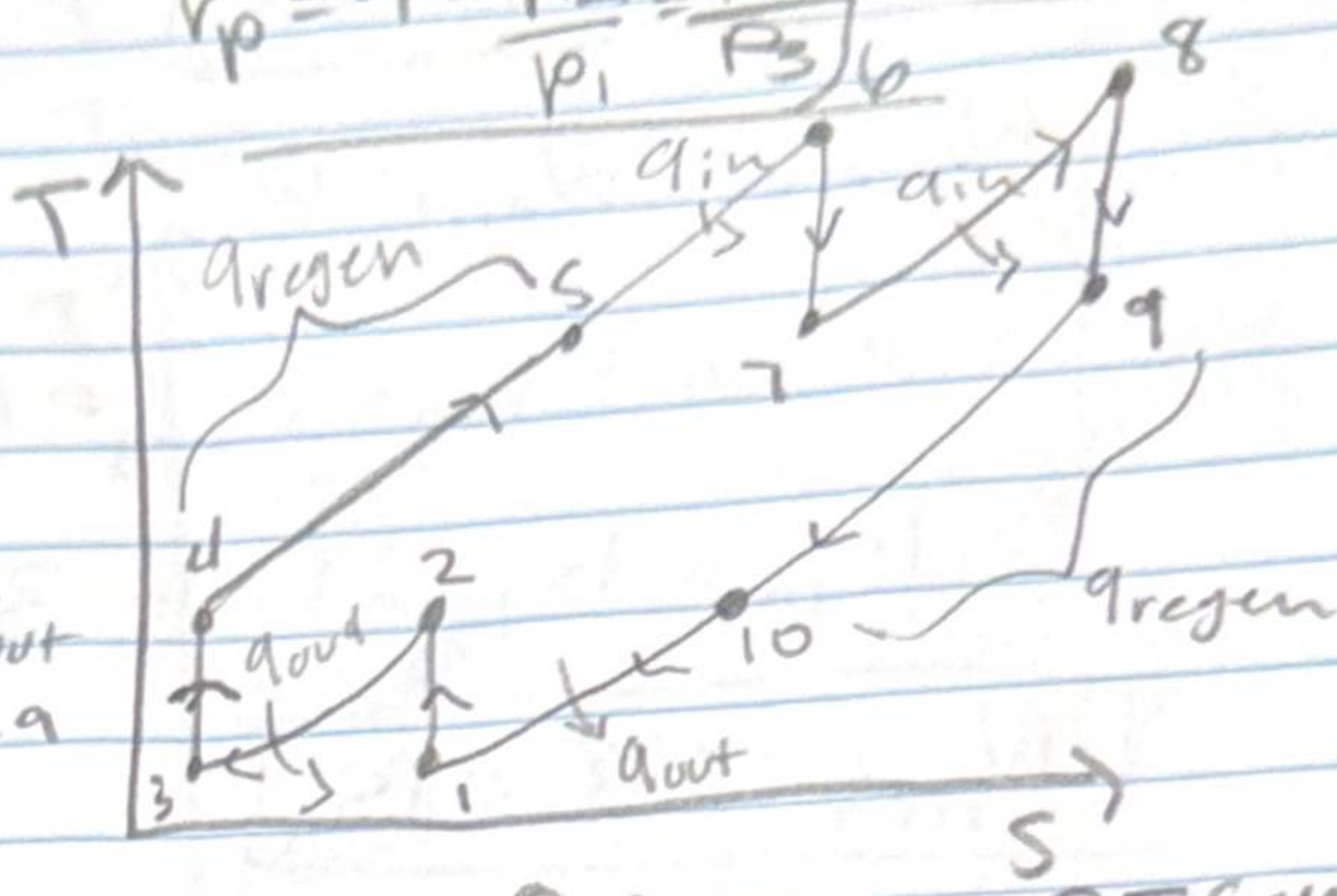
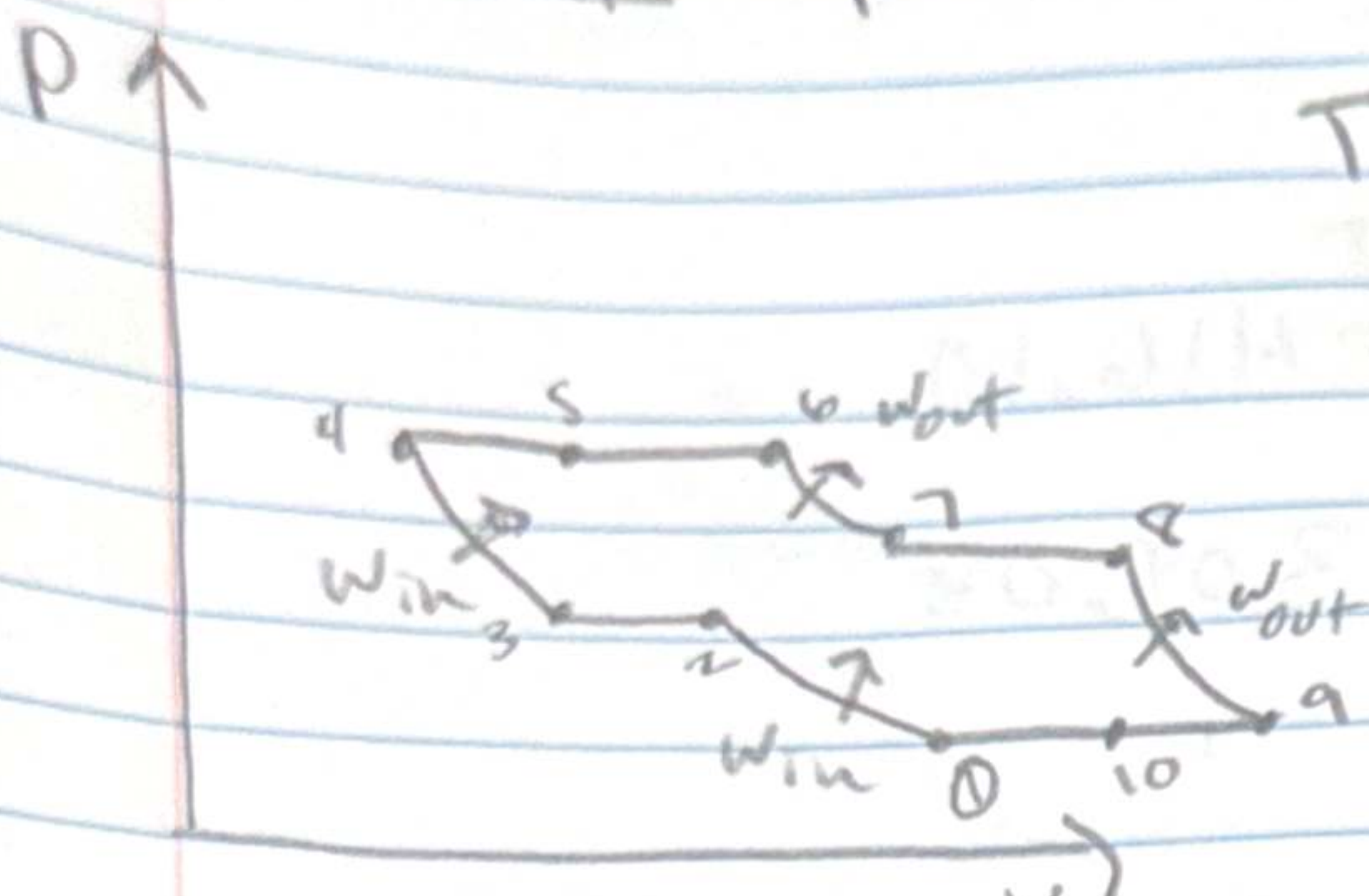
$$\boxed{= 0.225}$$

From A-17
 $h = 300.19$
 $P_r = 1.38$

$$\frac{P_2}{P_1} = \frac{P_4}{P_3} \quad \& \quad \frac{P_6}{P_7} = \frac{P_8}{P_9}$$

$$r_p = 4 = \frac{P_2}{P_1} = \frac{P_4}{P_3}$$

$$9-119 \quad r_p = 9$$



① Isen ② Const ③ Isent ④ P const reheat ⑤ P const reheat ⑥ Isent

$$T_1 = 300K \quad P_2 = 9P_1 \quad T_3 = 300K$$

$$T_6 = 1200K$$

$$P_2 = P_3$$

$$P_4 = P_8 = P_6$$

$$\frac{P_2}{P_1} = \frac{P_{r2}}{P_{r1}}$$

$$T_4 = T_2 = T_{10}$$

$$T_5 = T_7 = T_9$$

① Isentropic ② P const reheat ③ Isent ④ P const reheat ⑤ P const

$$T_9 = 1200K$$

$$P_7 = P_8$$

$$P_4 = P_{10} = P_1$$

* Confusion on Solving for Some Stages *