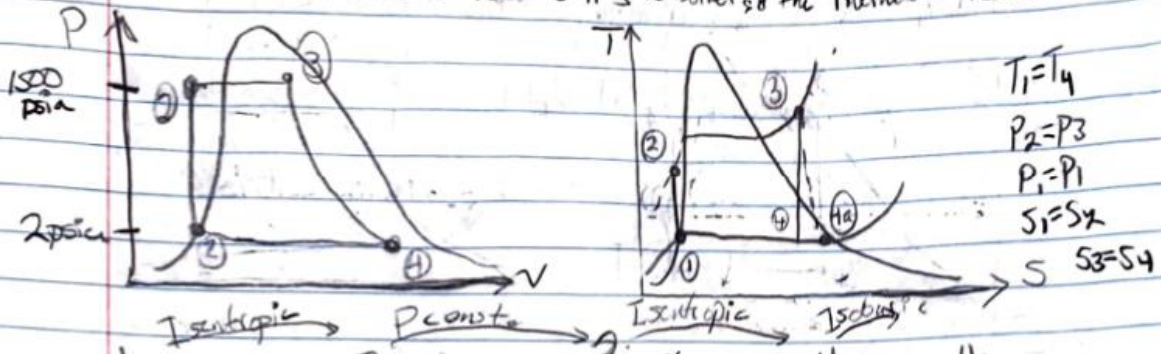


10-18E Steam Rankine cycle operates betw/ pressure limits of 1500 psia in the boiler & 2 psia in condenser. Turbine inlet temp is 800°F. Turbine isentropic efficiency is 90%. Ideal, cycle is sized to produce 2500 kW of power. Calculate mass flow rate through boiler, the power to produce by turbines, the rate of heat supply in boiler, & the Thermal efficiencies



$T_1 = T_4$
 $P_2 = P_3$
 $P_1 = P_4$
 $S_1 = S_2$
 $S_3 = S_4$

	1	2	3	4	4a
psia	2	1500	1500	2	
°F	126.02		800	126.02	
S_{table} (Btu/lbm-R)	0.17499	0.17499	1.5064	1.5064	
x	0			0.76	
h (Btu/lbm)	94.02	118.33	1363.1	803.81	859.79

$h_1 = h_f @ P_1 \Rightarrow 94.02 \text{ Btu/lbm}$
 $S_1 = S_f @ P_1 = 0.17499 \text{ Btu/lbm-R}$
 $V_1 = V @ P_1 = 0.01623 \text{ ft}^3/\text{lbm}$
 $w_{in} = v(P_2 - P_1) = h_2 - h_1$
 $h_2 = h_1 + v_1(P_2 - P_1) \Rightarrow 118.33 \text{ Btu/lbm}$
 $q_{in} = h_3 - h_2 \Rightarrow 1244.77 \text{ Btu/lbm}$
 $q_{out} = h_4 - h_1 \Rightarrow 859.79 - 94.02 \Rightarrow 765.77 \text{ Btu/lbm}$
 $w_{net} = q_{in} - q_{out} \Rightarrow 479.054 \text{ Btu/lbm}$
 $\eta_T = \frac{h_3 - h_4}{h_3 - h_2} \Rightarrow \frac{1363.1 - 803.81}{1363.1 - 118.33} \Rightarrow 0.9$

$h_3 @ P_3 \Rightarrow 1363.1 \text{ Btu/lbm}$
 $S_3 @ P_3 \Rightarrow 1.5064 \text{ Btu/lbm-R}$
 $S_f @ P_4 = 0.17499$ $S_{fg} @ P_4 = 1.74444$
 $x_4 = \frac{S_4 - S_f}{S_{fg}} = \frac{1.5064 - 0.17499}{1.74444} \Rightarrow x_4 = 0.76$
 $h_f @ P_4 = 94.02$
 $h_{fg} @ P_4 = 1021.7$
 $h_4 = h_f + x_4(h_{fg}) \Rightarrow h_4 = 803.81 \text{ Btu/lbm}$

i) Mass Flow rate through boiler (2-3):

$\dot{m} = \frac{W}{w_{net}} \Rightarrow \frac{2500 \text{ kW}}{479.054 \text{ Btu/lbm}} \times \frac{1 \text{ Btu}}{1055 \text{ J}} = 4.94 \frac{\text{lbm}}{\text{s}} \text{ or } 5 \frac{\text{lbm}}{\text{s}} = \dot{m}$

b.) Power produced by Turbine (3-4):

$$W_T = h_3 - h_4 \Rightarrow 1368.1 - 859.7398$$

$$W_T = 503.361$$

10-18E
Power $\dot{W}_T = \dot{m}(h_3 - h_4)$

$$\dot{W}_T = 5 \frac{\text{lbm}}{\text{s}} (503.361) \frac{\text{Btu}}{\text{lbm}} \Rightarrow 2516.805 \frac{\text{Btu}}{\text{s}} = \dot{W}_T$$

c.) Rate of heat supply (\dot{Q}_{in}) in boiler (2-3):

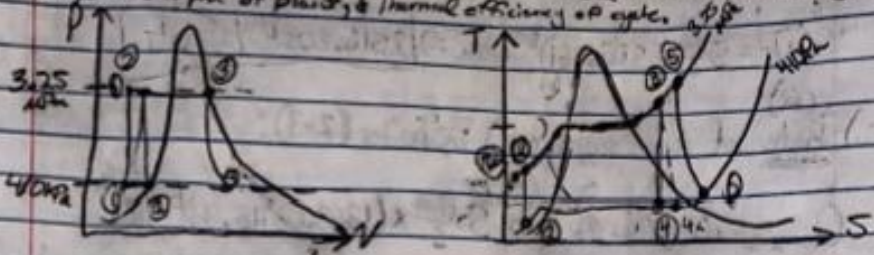
$$\dot{Q}_{in,2-3} = \dot{m}(h_3 - h_2) \Rightarrow 5 \frac{\text{lbm}}{\text{s}} \times (1363.1 - 118.33) \frac{\text{Btu}}{\text{lbm}}$$

$$\dot{Q}_{2-3} = 6223.85 \frac{\text{Btu}}{\text{s}}$$

d.) Thermal Efficiency: $\eta_{TH} = 1 - \frac{q_{out}}{q_{in}}$

$$\eta_{TH} = 1 - \frac{765.716}{1244.77} \Rightarrow 0.385 \text{ or } 38.5\% = \eta_{TH}$$

10-25 Binary geothermal power plant uses geothermal water @ 160°C as heat source. Cycle operates on simple Rankine cycle with isobutane as working fluid. Heat is transferred to cycle by heat exchanger, geothermal water enters @ 160°C @ Rate 555.9 kg/s & leaves @ 90°C. Isobutane enters turbine @ 3.25 MPa @ 147°C @ Rate 305.6 kg/s, leaves @ 79.5°C @ 410 kPa. Assume pump isentropic efficiency of 90%. Find isentropic efficiency of turbine, net power output of plant, & Thermal efficiency of cycle.



	1	2	3	4	5	6
°C	160	90	147	79.5	160	90
kPa	410	3.25	3.25	410	160	90
m ³ /s	555.9	555.9	305.6			

Isobutane $s_1 = s_2$ $h_4 @ P_4 = 608.356 \text{ kJ/kg}$ Isobutane

$h_1 = 608.356$ $s_2 = 1.785 \text{ kJ/kg}\cdot\text{K}$ $s_3 @ P_3 = 1.785 \text{ kJ/kg}\cdot\text{K}$ $P_2 @ T_2 = 70.183$

$s_1 = 1.785$ $h_2 - h_1 = v_1(P_2 - P_1)$ $x_{4a} = \frac{s_{4a} - s_3}{s_{4a} - s_3}$ $h_4 @ T_4 = 377.04$

$v_1 = 0.002$ $P_2 = 611.2 \text{ kPa}$ $s_{4a} = 1.1929$

$x_{4a} = 0.436$

$h_3 @ P_3, T_3 = 761.54 \text{ kJ/kg}$ $h_{4a} = h_3 \cdot x_{4a} + h_{4a} \cdot (1 - x_{4a})$

$s_{3a} = 2.5467$ $h_{4a} = 1537.297$

$h_{4a} @ s_{4a} = 608.226$ $h_{4a} @ s_{4a} = 2130.847$

a) Isentropic efficiency of turbine: $\eta_T = \frac{h_3 - h_{4a}}{h_3 - h_4}$

$\eta_T = \frac{761.54 - 1537.297}{761.54 - 608.356}$

$\eta_T = -5.0642$ I don't know what's wrong

b) Net power of plant

$W = \dot{m}_3 (w_{turb})$ $w_{turb} = (h_3 - h_4) - (h_2 - h_1)$

$w_{turb} = 150.34 \text{ kJ/kg}$

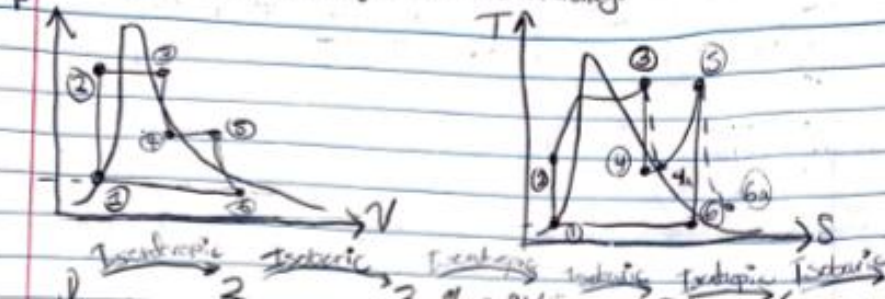
$W_{net} = 45945.13 \text{ kW}$

$\approx 46 \text{ MW}$

c) Thermal efficiency: $\eta_{th} = \frac{150.34 \text{ kJ/kg}}{150.34 \text{ kJ/kg}} = 1.0$ or 100%

Not ideal $q_{in} = h_3 - h_2$ $q_{out} = 150.34$

10-3) Steam Power plant operating as ideal reheat Rankine cycle.
 Boiler @ 5,000 kPa, reheat section @ 1200 kPa and condenser 20 kPa.
 Moisture quality at exit of both turbines = 5.96%. Find temperature
 inlet of each turbine and cycle thermal efficiency.

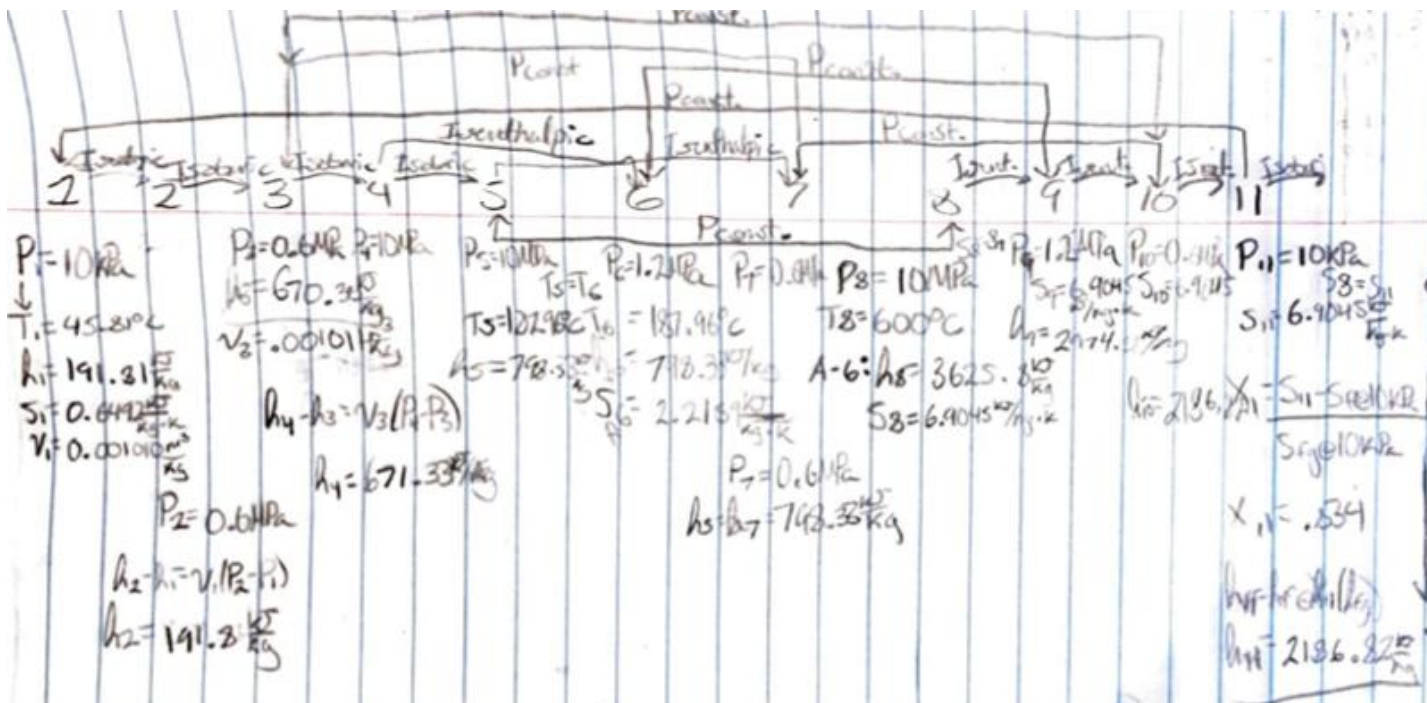


	1	2	3	4	5	6
Pressure (kPa)	100	5,000	5,000	1,200	1,200	100
Temperature (°C)				187.96		99.61
Quality (x)	0			0.96		0.96
Enthalpy h (kJ/kg)	419.17	424.28	3027.325	2777.014	3071.643	2584.806
Entropy s (kJ/kg·K)	1.3072	1.3072	6.3813	6.3813	7.116752	7.116752

$P_1 = P_6$ $S_1 = S_2$ $P_2 = P_3$ $S_3 = S_4$ $P_4 = P_5$ $S_5 = S_6$ $T_{6@P_6} = 99.61^\circ\text{C}$
 $h_{1@P_1} = 419.17 \text{ kJ/kg}$ $(A-S)$ $P_3 = 5000 \text{ kPa}$ $T_{4@P_4} = 187.96^\circ\text{C}$ $S_{6@P_6} = 7.116752 \text{ kJ/kg}\cdot\text{K}$
 $S_{1@P_1} = 1.3072 \text{ kJ/kg}\cdot\text{K}$ $h_3 = 3027.325 \text{ kJ/kg}$ $S_4 = S_3 + x_4(s_g - s_f)$ $S_{5@P_5} = 7.116752 \text{ kJ/kg}\cdot\text{K}$
 $x_4 = 0.001043$ $T_3 = 335.335^\circ\text{C}$ $S_4 = 6.38126 \text{ kJ/kg}\cdot\text{K}$ $S_6 = S_5 + x_6(s_g - s_f)$
 $x_6 = 0$ $h_4 = h_3 + x_4(h_g - h_f)$ $h_4 = 2777.014 \text{ kJ/kg}$ $h_6 = h_{f6} + x_6(h_{g6} - h_{f6})$
 $h_2 = v_f(P_2 - P_1) + h_1$ $h_2 = 424.28 \text{ kJ/kg}$ $h_5 = 3071.643 \text{ kJ/kg}$
 $h_3 = 3027.325 \text{ kJ/kg}$ $h_4 = 2777.014 \text{ kJ/kg}$ $h_6 = 2584.806 \text{ kJ/kg}$
 6.2111 300 $P_5 = 1200 \text{ kPa}$ $(A-S)$
 6.4516 350 $S_6 = 7.116752 \text{ kJ/kg}\cdot\text{K}$
 6.3813 $h_5 = 3071.643 \text{ kJ/kg}$

a) Temperature inlet of each turbine (4) (6):
 $T_3 = 335.335^\circ\text{C}$ $T_5 = 323.074^\circ\text{C}$

b) Thermal efficiency: $W_{net} = W_{out} + W_{out, 5-6} - W_{in, 1-2}$
 $\eta_{th} = \frac{W_{net}}{Q_{in}} = 0.253$ or 25.3%
 $Q_{in} = q_{in, 2-3} + q_{in, 4-5}$
 $Q_{in} = (h_3 - h_2) + (h_5 - h_4) = 732.038 \text{ kJ/kg}$
 $W_{net} = (h_3 - h_4) + (h_5 - h_6) - (h_2 - h_1) = 732.038 \text{ kJ/kg}$
 $Q_{in} = (h_3 - h_2) + (h_5 - h_4) = 2297.105 \text{ kJ/kg}$



$$y \times h_9 - y \times h_6 = h_5 - h_4 = y(h_9 - h_6) = (h_5 - h_4) \Rightarrow y = 0.05236$$

$$z = y \times h_7 + z \times h_{10} = h_3 \Rightarrow z = \frac{h_3 - y \times h_7}{h_{10}} \Rightarrow z = 0.285$$

$$(1 - y - z) \Rightarrow \text{out entropy}$$

$$q_{in} = h_8 - h_5 \Rightarrow q_{in} = 2877.47 \frac{\text{kJ}}{\text{kg}}$$

$$q_{out} = (1 - y - z)(h_{11} - h_1)$$

$$q_{out} = 1310 \frac{\text{kJ}}{\text{kg}}$$

10-53 pt3

a) Mass flow rate of steam, net power output 400 MW.

$$m = \frac{400 \text{ MW}}{w_{\text{net}}}$$

$\leftarrow 400000$

$$w_{\text{net}} = q_{\text{in}} - q_{\text{out}} = 1517.47 \frac{\text{kJ}}{\text{kg}}$$

$$m = 263.6 \frac{\text{kg}}{\text{s}}$$

b) Thermal efficiency:

$$\eta_{\text{th}} = \frac{w_{\text{net}}}{q_{\text{in}}} = \frac{1517.47}{2827.47} = 0.537 \text{ or } 53.7\%$$